

**10**  
YEARS

OF UNIVERSITY  
RECOGNITION

**20** YEARS OF  
ACADEMIC  
EXCELLENCE



**REVA**  
UNIVERSITY

Bengaluru, India

# **School of Applied Sciences** **M.Sc. PHYSICS**

## **HANDBOOK-2018-2020**

Rukmini Knowledge Park

Kattigenahalli, Yelahanka, Bengaluru – 560064

## **School of Applied Sciences**

### **M. Sc. (Physics)**

## **HANDBOOK**

**2018**

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## Chancellor's Message

*"Education is the most powerful weapon which you can use to change the world."*

- Nelson Mandela.

There was a time when survival depended on just the realization of physiological needs. We are indeed privileged to exist in a time when 'intellectual gratification' has become indispensable. Information is easily attainable for the soul that is curious enough to go look for it. Technological boons enable information availability anywhere anytime. The difference, however, lies between those who look for information and those who look for knowledge.



It is deemed virtuous to serve seekers of knowledge and as educators it is in the ethos at REVA University to empower every learner who chooses to enter our portals. Driven by our founding philosophy of 'Knowledge is Power', we believe in building a community of perpetual learners by enabling them to look beyond their abilities and achieve what they assumed impossible.

India has always been beheld as a brewing pot of unbelievable talent, acute intellect and immense potential. All it takes to turn those qualities into power is a spark of opportunity. Being at a University is an exciting and rewarding experience with opportunities to nurture abilities, challenge cognizance and gain competence.

For any University, the structure of excellence lies in the transitional abilities of its faculty and its facility. I'm always in awe of the efforts that our academic board puts in to develop the team of subject matter experts at REVA. My faculty colleagues understand our core vision of empowering our future generation to be ethically, morally and intellectually elite. They practice the art of teaching with a student-centered and transformational approach. The excellent infrastructure at the University, both educational and extra-curricular, magnificently demonstrates the importance of ambience in facilitating focused learning for our students.

A famous British politician and author from the 19th century - Benjamin Disraeli, once said 'A University should be a place of light, of liberty and of learning'. Centuries later this dictum still inspires me, and I believe it takes teamwork to build successful institutions. I welcome you to REVA University to join hands in laying the foundation of your future with values, wisdom, and knowledge.

**Dr. P. Shyama Raju**

The Founder and Hon'ble Chancellor, REVA University

## Vice-Chancellor's Message

The last two decades have seen a remarkable growth in higher education in India and across the globe. The move towards inter- disciplinary studies and interactive learning have opened up several options as well as created multiple challenges. India is at a juncture where a huge population of young crowd is opting for higher education. With the tremendous growth of privatization of education in India, the major focus is on creating a platform for quality in knowledge enhancement and bridging the gap between academia and industry.



A strong believer and practitioner of the dictum “Knowledge is Power”, REVA University has been on the path of delivering quality education by developing the young human resources on the foundation of ethical and moral values, while boosting their leadership qualities, research culture and innovative skills. Built on a sprawling 45 acres of green campus, this ‘temple of learning’ has excellent and state-of-the-art infrastructure facilities conducive to higher teaching-learning environment and research. The main objective of the University is to provide higher education of global standards and hence, all the programs are designed to meet international standards. Highly experienced and qualified faculty members, continuously engaged in the maintenance and enhancement of student- centric learning environment through innovative pedagogy, form the backbone of the University.

All the programs offered by REVA University follow the Choice Based Credit System (CBCS) with Outcome Based Approach. The flexibility in the curriculum has been designed with industry-specific goals in mind and the educator enjoys complete freedom to appropriate the syllabus by incorporating the latest knowledge and stimulating the creative minds of the students. Bench marked with the course of studies of various institutions of repute, our curriculum is extremely contemporary and is a culmination of efforts of great think-tanks - a large number of faculty members, experts from industries and research level organizations. The evaluation mechanism employs continuous assessment with grade point averages. We believe sincerely that it will meet the aspirations of all stakeholders – students, parents and the employers of the graduates and postgraduates of Reva University.

At REVA University, research, consultancy, and innovation are regarded as our pillars of success. Most of the faculty members of the University are involved in research by attracting funded projects from various research level organizations like DST, VGST, DBT, DRDO, AICTE and industries. The outcome of the research is passed on to students through live projects from industries. The entrepreneurial zeal of the students is encouraged and nurtured through EDPs and EACs.

REVA University has entered into collaboration with many prominent industries to bridge the gap between industry and University. Regular visits to industries and mandatory internship with industries have helped our students. REVA University has entered into collaboration with many prominent industries to bridge the gap between industry and University. Regular visits to industries and mandatory internships with industries have helped our students become skilled with relevant to industry requirements. Structured training programs on soft-skills and preparatory training for competitive exams are offered here to make students more employable. 100% placement of eligible students speaks of the effectiveness of these programs. The entrepreneurship development activities and establishment of “Technology Incubation Centers” in the University extend full support to the budding entrepreneurs to nurture their ideas and establish an enterprise.

With firm faith in the saying, “Intelligence plus character –that is the goal of education” (Martin Luther King, Jr.), I strongly believe REVA University is marching ahead in the right direction, providing a holistic education to the future generation and playing a positive role in nation building. We reiterate our endeavor to provide premium quality education accessible to all and an environment for the growth of over-all personality development leading to generating “GLOBAL PROFESSIONALS”.

Welcome to the portals of REVA University!

**Dr. S. Y. Kulkarni**

Vice-Chancellor, REVA University

## Director Message

Physics is a basic science which enables the students to think beyond their comfort level and brings new and deeper concepts about the Atoms, Molecules, Nanoscale, Microscale and Bulk materials physical properties. The present-day buzzing “quantum computers” also relies on the principles of physics like Quantum Mechanics, Electrodynamics and Optics. Physical sciences deal with subatomic particles to Galaxies. It also gave birth to high end sophisticated technologies like Atomic Force microscope, Tunneling Electron microscope and other advanced technologies which help the people to visualize the unexplored world. M Sc in Physics offered by REVA University aims to provide the required skills and knowledge necessary to pursue a successful career in Research, Teaching, and Industry relevant jobs. This program imparts need based, practical education in contemporary world to develop global competence among students. It strives to prepare students to become leaders in the field of Physical Sciences in general and Material science/Optics/Electronics in particular by encouraging them to inculcate scientific thinking coupled with creative and innovative ideas.

The program provides hands- on training and practical skills in the field of Material Science, Optics and Electronics. This course also covers fundamental aspects of Physics in every aspect. The special emphasis on experiments related electricity, magnetism, lasers, spectral analysis and studying other materials properties by varying different parameters help the student to equip with required knowledge in the field of research, teaching & industry.

As far as employment is concerned physics courses are gaining attention in the field of data science, data analysis, statistical modelling and programming are some of the fast-growing sectors. Employment record shows that physics has a great scope in future. Physicists can find careers with electronic, optics and material science related and allied companies. They can be employed in the areas of programming, production and management of sensors and other related industries. There is a large-scale employment in research laboratories run by the government as well as the corporate sectors. Further, there is great demand for physicists in numerous industries and sectors after the completion of MSc Physics course.

This handbook provides an outline of regulations for master’s degree, scheme of instruction, and detailed syllabus. I am sure the students choosing MSc Physics at REVA University will enjoy the curriculum, teaching and learning environment, the vast infrastructure and the experienced teacher’s involvement and guidance. We will strive to provide all the necessary comfort and congenial environment for their studies. I wish all students a pleasant stay at REVA and grand success in their career.

The curriculum caters to and has relevance to local, regional, national, global developmental needs.

The maximum number of courses are integrated with cross cutting issues relevant to professional ethics, gender, human values, environment, and sustainability.

Dr. Beena G  
Director  
School of Applied Sciences

## PREFACE

Higher education across the globe is opening doors of its academic disciplines to real-world experiences. The disciplinary legitimacy is under critical review. Trans-border mobility and practice learning are being foregrounded as guiding principles. Interactive learning, bridging disciplines and facilitating learners to gain different competencies through judicious management of time is viewed as one of the greatest and fascinating priorities and challenges today.

The M.Sc. in Physics is designed keeping in view the current situation and possible future developments, both at national and global levels. This course is designed to give greater emphasis on Research. There are an ample number of courses providing knowledge in specialized areas of Quantum Mechanics, Electrodynamics, Electronics, Materials Science, etc. Facilitating students to choose specialized areas of their interest. Adequate attention is given to providing students with the basic concepts of analysis and modern computation techniques to be used and knowledge on application of such concepts in practical field. The project, being part of the curriculum, will certainly provide students the research experience.

The L: T: P structure of teaching and learning under Choice Based Credit System (CBCS) and Continuous Assessment Grading Pattern (CAGP) would certainly help our students learn and build competencies needed in this knowledge-based society.

This handy document containing brief information about M.Sc. Physics, scheme of instruction, course content, CBCS-CAGP regulations and its advantages and calendar of events for the year will serve as a guiding path to students to move forward in a right direction. It would mould them with knowledge, skill and ethical values to face the challenges of this competitive world with greater confidence in becoming proud citizens of Mother India.

The curriculum caters to and has relevance to local, regional, national, global developmental needs.

The maximum number of courses are integrated with cross cutting issues relevant to professional ethics, gender, human values, environment, and sustainability.

**Dr D V Sunitha**

Coordinator, M Sc Physics Program

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## **RUKMINI EDUCATIONAL CHARITABLE TRUST**

It was the dream of late Smt. Rukmini Shyama Raju to impart education to millions of underprivileged children as she knew the importance of education in the contemporary society. The dream of Smt. Rukmini Shyama Raju came true with the establishment of Rukmini Educational Charitable Trust (RECT), in the year 2002. Rukmini Educational Charitable Trust (RECT) is a Public Charitable Trust, set up in 2002 with the objective of promoting, establishing and conducting academic activities in the fields of Arts, Architecture, Commerce, Education, Engineering, Environmental Science, Legal Studies, Management and Science & Technology, among others. In furtherance of these objectives, the Trust has set up the REVA Group of Educational Institutions comprising of REVA Institute of Technology & Management (RITM), REVA Institute of Science and Management (RISM), REVA Institute of Management Studies (RIMS), REVA Institute of Education (RIE), REVA First Grade College (RFGC), REVA Independent PU College at Kattigenahalli, Ganganagar and Sanjaynagar and now REVA University. Through these institutions, the Trust seeks to fulfill its vision of providing world class education and create abundant opportunities for the youth of this nation to excel in the areas of Arts, Architecture, Commerce, Education, Engineering, Environmental Science, Legal Studies, Management and Science & Technology.

Every great human enterprise is powered by the vision of one or more extraordinary individuals and is sustained by the people who derive their motivation from the founders. The Chairman of the Trust is Dr. P. Shyama Raju, a developer and builder of repute, a captain of the industry in his own right and the Chairman and Managing Director of the DivyaSree Group of companies. The idea of creating these top notched educational institutions was born of the philanthropic instincts of Dr. P. Shyama Raju to do public good, quite in keeping with his support to other socially relevant charities such as maintaining the Richmond road park, building and donating a police station, gifting assets to organizations providing accident and trauma care, to name a few.

The Rukmini Educational Charitable Trust drives with the main aim to help students who are in pursuit of quality education for life. REVA is today a family of ten institutions providing education from PU to Post Graduation and Research leading to PhD degrees. REVA has well qualified experienced teaching faculty of whom majority are doctorates. The faculty is supported by committed administrative and technical staff. Over 11,000 students study various courses across REVA's three campuses equipped with exemplary state-of-the-art infrastructure and conducive environment for the knowledge driven community.

## **ABOUT REVA UNIVERSITY**

REVA University has been established under the REVA University Act, 2012 of Government of Karnataka and notified in Karnataka State Gazette dated 7<sup>th</sup> February, 2013. The University is empowered by UGC to award degrees to any branch of knowledge under Sec.22 of the UGC Act. The University is a Member of Association of Indian Universities, New Delhi. The main objective of the University is to prepare students with knowledge, wisdom, and patriotism to face global challenges and become the top leaders of the country and the globe in different fields.

REVA University located in between Kempegowda International Airport and Bangalore city, has a sprawling green campus spread over 45 acres of land and equipped with state-of-the-art infrastructure that provide conducive environment for higher learning and research. The REVA campus has well equipped laboratories, custom-built teaching facilities, fully air-conditioned library and central computer center, the well-planned sports facility with cricket ground, running track & variety of indoor and outdoor sports activities, facilities for cultural programs. The unique feature of REVA campus is the largest residential facility for students, faculty members and supportive staff.

The University is presently offering 23 Post Graduate Degree programs, 20 Degree and PG Degree programs in various branches of studies and has 12000+ students studying in various branches of knowledge at graduate and post graduate level and 302 Scholars pursuing research leading to PhD in 18 disciplines. It has 800+ well qualified, experienced, and committed faculty members of whom the majority are doctorates in their respective areas and most of them are guiding students pursuing research leading to PhD.

The programs being offered by the REVA University are well planned and designed after detailed study with emphasis on knowledge assimilation, applications, global job market and their social relevance. Highly qualified, experienced faculty and scholars from reputed universities / institutions, experts from industries and business sectors have contributed to preparing the scheme of instruction and detailed curricula for this program. Greater emphasis on practice in respective areas and skill development to suit to respective job environment has been given while designing the curricula. The Choice Based Credit System and Continuous Assessment Graded Pattern (CBCS – CAGP) of education has been introduced in all programs to facilitate students to opt for subjects of their choice in addition to the core subjects of the study and prepare them with needed skills. The system also allows students to move forward under the fast track for those who have the capabilities to surpass others. These programs are taught by well experienced qualified faculty supported by the experts from industries,

business sectors and such other organizations. REVA University has also initiated many supportive measures such as bridge courses, special coaching, remedial classes, etc., for slow learners so as to give them the needed input and build in them confidence and courage to move forward and accomplish success in their career. The University has also entered into MOUs with many industries, business firms and other institutions seeking their help in imparting quality education through practice, internship and also assisting students' placements.

REVA University, recognizing the fact that research, development, and innovation are the important functions of any university, has established an independent Research and Innovation division headed by a senior professor as Dean of Research and Innovation. This division facilitates all faculty members and research scholars to undertake innovative research projects in engineering, science & technology and other areas of study. Interdisciplinary-multidisciplinary research is given the topmost priority. The division continuously liaisons between various funding agencies, R&D Institutions, Industries and faculty members of REVA University to facilitate undertaking innovative projects. It encourages student research projects by forming different research groups under the guidance of senior faculty members. Some of the core areas of research wherein our young faculty members are working include Data Mining, Cloud Computing, Image Processing, Network Security, VLSI and Embedded Systems, Wireless Sensor Networks, Computer Networks, IOT, MEMS, Nano- Electronics, Wireless Communications, Bio- fuels, Nano-technology for coatings, Composites, Vibration Energies, Electric Vehicles, Multilevel Inverter Application, Battery Management System, LED Lightings, Renewable Energy Sources and Active Filter, Innovative Concrete Reinforcement, Electro Chemical Synthesis, Energy Conversion Devices, Nano-structural Materials, Photo-electrochemical Hydrogen generation, Pesticide Residue Analysis, Nano materials, Photonics, Nano Tribology, Fuel Mechanics, Operation Research, Graph theory, Strategic Leadership and Innovative Entrepreneurship, Functional Development Management, Resource Management and Sustainable Development, Cyber Security, General Studies, Feminism, Computer Assisted Language Teaching, Culture Studies etc.

REVA University has also given utmost importance to developing the much-required skills through a variety of training programs, industrial practice, case studies and such other activities that induce the said skills among all students. A full-fledged Career Development and Placement (CDC) department with world class infrastructure, headed by a dynamic experienced Professor & Dean, and supported by well experienced Trainers, Counselors and Placement Officers. The University also has University-Industry Interaction and Skill Development Centre headed by a Senior Professor & Director facilitating skill related training to REVA students and other unemployed students. The University has been recognized as a Centre of Skill Development and Training by NSDC (National Skill Development Corporation) under Pradhan Mantri Kaushal Vikas Yojana. The Centre conducts several add-on courses

in challenging areas of development. It is always active in facilitating student's variety of Skill Development Training programs.

The University has collaborations with Industries, universities abroad, research institutions, corporate training organizations, and Government agencies such as Florida International University, Oklahoma State University, Western Connecticut University, University of Alabama, Huntsville, Oracle India Ltd, Texas Instruments, Nokia University Relations, EMC<sup>2</sup>, VMware, SAP, Apollo etc, to facilitate student exchange and teacher-scholar exchange programs and conduct training programs. These collaborations with foreign universities also facilitate students to study some of the programs partly in REVA University and partly in foreign university, viz, M.S in Computer Science one year in REVA University and the next year in the University of Alabama, Huntsville, USA.

The University has also given greater importance to quality in education, research, administration, and all activities of the university. Therefore, it has established an independent Internal Quality division headed by a senior professor as Dean of Internal Quality. The division works on planning, designing and developing different quality tools, implementing them and monitoring the implementation of these quality tools. It concentrates on training the entire faculty to adopt the new tools and implement their use. The division further works on introducing various examination and administrative reforms.

To motivate the youth and transform them to become innovative entrepreneurs, successful leaders of tomorrow and committed citizens of the country, REVA organizes interaction between students and successful industrialists, entrepreneurs, scientists, and such others from time to time. As a part of this exercise great personalities such as Bharat Ratna Prof. C. N. R. Rao, a renowned Scientist, Dr. N R Narayana Murthy, Founder and Chairman and Mentor of Infosys, Dr. K Kasturirangan, Former Chairman ISRO, Member of Planning Commission, Government of India, Dr. Balaram, Former Director I.I.Sc., and noted Scientist, Dr. V S Ramamurthy, Former Secretary, DST, Government of India, Dr. V K Aatre, noted Scientist and former head of the DRDO and Scientific Advisor to the Ministry of Defense Dr. Sathish Reddy, Scientific Advisor, Ministry of Defense, New Delhi and many others have accepted our invitation and blessed our students and faculty members by their inspiring addresses and interaction.

As a part of our effort in motivating and inspiring youth of today, REVA University also has instituted awards and prizes to recognize the services of teachers, researchers, scientists, entrepreneurs, social workers and such others who have contributed richly for the development of the society and progress of the country. One of such awards instituted by REVA University is '**Lifetime Achievement Award**' to be awarded to successful personalities who have made mark in their field of work. This award is presented on occasion of the "**Founders' Day Celebration**" of REVA University on 6<sup>th</sup> January of every year in presence of dignitaries, faculty members and students gathering. The first "REVA Lifetime

Achievement Award” for the year 2015 has been awarded to Shri. Kiran Kumar, Chairman ISRO, followed by Shri. Shekhar Gupta, renowned Journalist for the year 2016, Dr K J Yesudas, renowned play back singer for the year 2017. REVA also introduced “**REVA Award of Excellence**” in the year 2017 and the first Awardee of this prestigious award is Shri Ramesh Aravind, Actor, Producer, Director, Screen Writer and Speaker.

REVA organizes various cultural programs to promote culture, tradition, ethical and moral values to our students. During such cultural events the students are given opportunities to unfold their hidden talents and motivate them to contribute innovative ideas for the progress of society. One of such cultural events is REVAMP conducted every year. The event not only gives opportunities to students of REVA but also students of other Universities and Colleges. During three days of this mega event students participate in debates, Quizzes, Group discussion, Seminars, exhibitions, and variety of cultural events. Another important event is Shubha Vidaaya, - Graduation Day for the final year students of all the programs, wherein, the outgoing students are felicitated and are addressed by eminent personalities to take their future career in a right spirit, to be the good citizens and dedicate themselves to serve the society and make a mark in their respective spheres of activities. During this occasion, the students who have achieved top ranks and won medals and prizes in academic, cultural and sports activities are also recognized by distributing awards and prizes. The founders have also instituted medals and prizes for sports achievers every year. The physical education department conducts regular yoga classes every day to students, faculty members, administrative staff and their family members and organizes yoga camps for villagers around.

Recognizing the fast growth of the university and its quality in imparting higher education, the BERG (Business Excellence and Research Group), Singapore has awarded BERG Education Award 2015 to REVA University under Private Universities category. The University has also been honored with many more such honors and recognitions.

## **Vision**

REVA University aspires to become an innovative university by developing excellent human resources with leadership qualities, ethical and moral values, research culture and innovative skills through higher education of global standards

## **Mission**

- To create excellent infrastructure facilities and state-of-the-art laboratories and incubation centers
- To provide student-centric learning environment through innovative pedagogy and education reforms
- To encourage research and entrepreneurship through collaborations and extension activities
- To promote industry-institute partnerships and share knowledge for innovation and development
- To organize society development programs for knowledge enhancement in thrust areas
- To enhance leadership qualities among the youth and enrich personality traits, promote patriotism and moral values.

## **Objectives**

- Creation, preservation and dissemination of knowledge and attainment of excellence in different disciplines
- Smooth transition from teacher - centric focus to learner - centric processes and activities
- Performing all the functions of interest to its major constituents like faculty, staff, students and the society to reach leadership position
- Developing a sense of ethics in the University and Community, making it conscious of its obligations to the society and the nation
- Accepting the challenges of globalization to offer high quality education and other services in a competitive manner

## **ABOUT SCHOOL OF APPLIED SCIENCES**

The School of Applied Sciences offers graduate and post graduate programs in Biotechnology, Biochemistry, Chemistry, Physics and Mathematics which are incredibly fascinating. It aims to attract talented youth and train them to acquire knowledge and skills useful to industrial sectors, research laboratories, and educational institutions. The school presently offers M.Sc. degree programs in Bio-Chemistry, Bio-Technology, Chemistry, Physics, Mathematics and B Sc with various combinations viz, Physics Chemistry and Mathematics, Mathematics, Physics and Statistics, Mathematics Statistics and Computer Science, and Biology Mathematics & Computer Science and also Post Graduate Diploma in Clinical Research Management. The school also facilitates research leading to PhD in Biotechnology, Biochemistry, Physics, Chemistry, Mathematics and related areas of study.

The School of Applied Sciences is shouldered by well qualified, experienced and highly committed faculty. The state-of-the-art infrastructure digital classrooms, well equipped laboratories, conference rooms and the serene academic atmosphere at REVA University will enhance the transfer as well as creation of knowledge. The school provides an interactive, collaborative peer tutoring environment that encourages students to break down complex problems and develop strategies for finding solutions across a variety of situations and disciplines. The school aims to develop a learning community of critical thinkers who serves as models of innovative problems solving in the university environment to enrich their academic and professional careers.

### **Vision**

To nurture intellect, creativity, character and professionalism among students and impart contemporary knowledge in various branches of Chemical, Biological, Physical and Mathematical Sciences that are socially relevant and transform them to become global citizens.

### **Mission**

To achieve excellence in studies and research through pedagogy and support interface between industry and academia

To create intellectual curiosity, academic excellence, and integrity through multidimensional exposure

To establish state of the art laboratories to support research and innovation and promote mastery of science.

To inculcate an ethical attitude and make students competitive to serve the society and nation.

## VALUES

- Excellence in all our academic and research endeavour
- Dedication and service to our stakeholders
- Leadership through innovation
- Accountability and transparency
- Creating conducive academic environment with service motto
- Integrity and intellectual honesty
- Ethical and moral behaviour
- Freedom of thought and expression
- Adaptability to the change
- Team-work

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“The constant questioning of our values and achievements is a challenge without which neither science nor society can remain healthy”

— **Aage Niels Bohr**



## BOS MEMBERS

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## **M. Sc. (Physics)**

### **Programme Overview**

Physics is a natural science. It deals with physical matter and energy; and the natural laws that govern the behavior of matter. The core theories of Physics are: Classical Mechanics, Electromagnetism, Thermodynamics and Statistical Mechanics, Quantum Mechanics and Relativity. There are many more branches of Physics including astronomy.

Physics plays a key role in the future progress of humankind. The physics education and research in all countries is important because:

1. Physics is an exciting intellectual adventure that inspires young people and expands the frontiers of our knowledge about Nature.
2. Physics generates fundamental knowledge needed for the future technological advances that will continue to drive the economic engines of the world.
3. Physics contributes to the technological infrastructure and provides trained personnel needed to take advantage of scientific advances and discoveries.
4. Physics is an important element in the education of chemists, engineers and computer scientists, as well as practitioners of the other physical and biomedical sciences.
5. Physics extends and enhances our understanding of other disciplines, such as the earth, agricultural, chemical, biological, and environmental sciences, plus astrophysics and cosmology - subjects of substantial importance to all peoples of the world.
6. Physics improves our quality of life by providing the basic understanding necessary for developing new instrumentation and techniques for medical applications, such as computer tomography, magnetic resonance imaging, positron emission tomography, ultrasonic imaging, and laser surgery.

Thus, physics is an essential part of the educational system of an advanced society. Indian Society has embraced knowledge economy and its economic growth rate is one of the highest in the world. India has shown highest level of progress in engineering, space, nuclear, aeronautics and information and communication technologies. The subject of physics has played a major role in the development of country and India has produced 2 Nobel laureates in Physics.

In this context, University across the country offer Physics as a subject at undergraduate and physics as a programme at postgraduate level.

M. Sc. (Physics) at REVA UNIVERSITY has been designed to meet the human resources needs of existing and futuristic research establishments, industries and academic institutions. The programme is designed to produce graduates with higher order critical, analytical, problem solving and research skills; ability to think rigorously and independently to meet higher level expectations of industries, research organization and academic institutions. The programme deals with courses in classical mechanics, quantum mechanics, material science, semiconductors, electrodynamics and related areas.

### **Eligibility for M Sc (Physics) program**

Passed Bachelor's Degree of 3 years with Physics as major / optional subject with 45% marks (40% in case of candidate belonging to SC/ST category) of marks in aggregate of any recognized / institution or any other qualification recognized as equivalent there to.

### **Programme Educational Objectives (PEOs)**

The aim of the programme is to produce postgraduates with - advanced knowledge and understanding of Physics; higher order critical, analytical, problem solving and attitudinal skills (transferable) to meet expectations of research establishments, relevant industry and academia or to take up entrepreneurial route. Hence,

#### **The Programme Educational objectives are to prepare the students to:**

**PEO-1** Pursue higher education through continuous learning with effective communication skills

**PEO-2** Have successful professional careers in academia, industry and government

**PEO-3** Start own enterprise and provide solutions to scientific research problems

**PEO-4** Exhibit skills as a member of a team in national and international organizations with highest ethics through lifelong learning

### **Programme Outcomes (POs)**

After undergoing this programme, a student will be able to:

**PO1. Domain knowledge:** Apply the knowledge of physics and fundamentals for the solution of complex problems in day to day life.

**PO2. Problem analysis:** Identify, formulate, research literature, and analyze problems to arrive at substantiated conclusions using principles of physical sciences.

**PO3. Design/development of solutions:** Design solutions for real time problems to meet the specifications with consideration for the public health and safety, the cultural and societal, and environmental considerations.

**PO4. Conduct investigations of complex problems:** Use research-based knowledge, for analysis and interpretation of data, and synthesis of the information to provide valid conclusions.

**PO5. Modern tool usage:** Apply appropriate techniques, resources, and IT tools including prediction and modeling to complex activities with an understanding of the limitations.

**PO6. Environmental and society:** Apply reasoning informed by the contextual knowledge to assess societal, health, safety, legal, and cultural issues and the consequent responsibilities relevant to the professional practice.

**PO7. Environment and sustainability:** Understand the impact of the solutions in societal and environmental contexts, and demonstrate the knowledge of, and need for sustainable development.

**PO8. Ethics:** Apply ethical principles and commit to ethics, and responsibilities and norms of the professional practice

**PO9. Individual and team work:** Function effectively as an individual, and as a member or leader in teams, and in multidisciplinary settings.

**PO10. Communication:** Communicate effectively with the professional community and with society at large. Be able to comprehend and write effective reports and documentation. Make effective presentations and give and receive clear instructions.

**PO11. Life-long learning:** Recognize the need for and have the preparation and ability to engage in independent and life-long learning in the broadest context of technological change.

## **PROGRAM SPECIFIC OUTCOMES**

**PSO1.** Apply the fundamentals of classical mechanics, electrodynamics, Quantum mechanics and condensed matter physics to understand the energy quantization concepts.

**PSO2.** Identify and compare the materials best suited for futuristic engineering applications

**PSO3.** Explore the knowledge of basic concepts of atomic, molecular, nuclear physics to analyse the spectra obtained from various bodies.

**PSO4.** Demonstrate the knowledge of fundamentals of electronic devices

## **REVA University Regulations for Choice Based Credit System (CBCS) and Continuous Assessment Grading Pattern (CAGP) for Postgraduate Degree programs- 2021-22**

(Framed as per the provisions under Section 35 (ii), Section 7 (x) and Section 8 (xvi) & (xxi) of the REVA University Act, 2012)

### **1. Tie and Commencement:**

1.1 These Regulations shall be called the “**REVA University Regulations for Choice Based Credit System (CBCS) and Continuous Assessment Grading Pattern (CAGP) for Post Graduate Degree Programs- 2020**”.

These Regulations shall come into force from the date of assent of the Chancellor.

### **1. The Programs:**

The following programs and all Graduate Degree programs to be instituted and introduced in REVA University in coming years shall follow these regulations.

**M.Sc. in:**

Biotechnology  
Biochemistry  
Chemistry  
Physics  
Mathematics  
Microbial Technology  
Bioinformatics

**2. Definitions:**

**Course:** Every course offered will have three components associated with the teaching-learning process of the course, namely:

(i) L= Lecture (ii) T= Tutorial (iii) P=Practice; where:

**L** stands for **Lecture** session consisting of classroom instruction.

**T** stands for **Tutorial** session consisting participatory discussion / self-study/ desk work/ brief seminar presentations by students and such other novel methods that make a student to absorb and assimilate more effectively the contents delivered in the Lecture classes.

**P** stands for **Practice** session and it consists of Hands on Experience / Laboratory Experiments / Field Studies / Case Studies that equip students to acquire the much required skill component.

**2.Courses of study and Credits**

4.1The study of various subjects in M. Sc., degree program is grouped under various courses. Each of these course carries credits which are based on the number of hours of teaching and learning.

4.1.1.In terms of credits, every **one-hour session of L amounts to 1 credit per Semester.**

In terms of credits, every **one-hour session of L amounts to 1 credit per Semester** and a minimum of **two-hour session of T or P amounts to 1 credit per Semester** over a period of one Semester of 16 weeks for teaching-learning process.

**4.1.1The total duration of a semester is 20 weeks inclusive of semester-end examination.**

**A course shall have either or all the four components.** That means a course may have only lecture component, or only practical component or combination of any two or all the three components.

**4.1.1The concerned BoS will assign Credit Pattern for every course based on the requirement. However, generally, courses can be assigned with 1-4 Credits depending on the size of the course.**

4.1.1Different **Courses of Study** are labelled and defined as follows:

**Core Course:**

A course which should compulsorily be studied by a candidate as a core-requirement is termed as a Core course. The CORE courses of Study are of THREE types, viz – (i) Hard Core Course, and (ii) Soft Core Course, (iii) Open Elective

**a. Hard Core Course (HC):**

The **Hard Core Course** is a Core Course in the main branch of study and related branch(es) of study, if any that the candidates have to complete compulsorily.

**a. Soft Core Course (SC):**

A Core course may be a **Soft Core** if there is a choice or an option for the candidate to choose a course from a pool of courses from the main branch of study or from a sister/related branch of study which supports the main branch of study.

**a. Open Elective Course:**

An elective course chosen generally from other discipline / related subject, with an intention to seek exposure to the related subjects other than the main discipline the student is studying is called an **Open Elective Course**.

**e. Project Work / Dissertation:**

Project work / Dissertation work is a special course involving application of knowledge in solving / analysing /exploring a real life situation / difficult problem. A project work carrying **FOUR or SIX** credits is called **Minor Project work / Dissertation**. A project work of **EIGHT, TEN, TWELVE or SIXTEEN** credits is called **Major Project work / Dissertation**. **A Project work may be a hard core or a Soft Core as decided by the BoS / concerned.**

**2. Eligibility for Admission:**

Bachelors Degree of three years with Biotechnology or any Life Science subject as one of the cognate / major / optional subjects with 60% (40% in case of candidates belonging to SC/ST) of marks in aggregate from any recognized University / Institution or any other qualification recognized as equivalent thereto.

**2. Scheme, Duration and Medium of Instructions:**

6.1. M.Sc., degree program is of 4 semesters - 2 years duration. A candidate can avail a maximum of 6 semesters (3 years) including blank semesters, if any to successfully complete M.Sc. degree. Whenever a candidate opts for blank semester, he/she has to study the prevailing courses offered by the School when he/she resumes his/her studies.

6.2. The medium of instruction shall be English.

**7. Credits and Credit Distribution**

7.1. A candidate has to earn 96 credits for successful completion of Two Year Postgraduate degree with a distribution of credits for different courses as given in Table - 1 given below:

**Table-1**

### Credits and Credit Distribution for Two Year Post Graduate degree program in Sciences

Course Type	Credits for Two Year (4 Semesters) Post Graduate Degree Programs
Hard Core Course	A minimum of 60 but not exceeding 70
Soft Core Course	A minimum of 10 but not exceeding 30
Open Elective	A minimum of 2
Project/Dissertation	A minimum of 10
MOOC / Swayam/ Coursera/Soft Skill Training	A minimum of 4
<b>Total</b>	<b>90</b>

- 7.2. The concerned BOS based on the credits distribution pattern given above shall prescribe the credits to various types of courses and shall assign title to every course including project work, practical work, field work, self-study elective, as **Hard Core (HC) or Soft Core (SC) Open Elective (OE) or Mandatory Course (MC)**
- 7.3. Every course including project work, practical work, field work, self-study elective should be entitled as Hard Core (HC) or Soft Core (SC) or Open Elective (OE) **or Mandatory Course (MC)** by the BoS concerned.
- 7.4. The concerned BOS shall specify the desired Program Objectives, Program Educational Objectives, Program Specific Outcomes and Course Outcomes while preparing the curriculum of a particular program.
- 7.5. A candidate can enrol for a maximum of 30 credits and a minimum of 20 credits per Semester. However he / she may not successfully earn a maximum of 30 credits per semester. This maximum of 30 credits does not include the credits of courses carried forward by a candidate.
- 7.6. Only such full time candidates who register for a minimum prescribed number of credits in each semester from I semester to IV semester and complete successfully 96 credits in 4 successive semesters shall be considered for declaration of Ranks, Medals, Prizes and are eligible to apply for Student Fellowship, Scholarship, Free ships, and such other rewards / advantages which could be applicable for all full time students and for hostel facilities.

### 8. Add-on Proficiency Certification / Diploma:

#### 8.1 Add- on Proficiency Certification:

To acquire Add on Proficiency Certification a candidate can opt to complete a minimum of 4 extra credits either in the same discipline /subject or in different discipline / subject in excess to 90 credits for the Two Year Post Graduate degree programs.

#### 8.2 Add on Proficiency Diploma:

To acquire Add on Proficiency Diploma, a candidate can opt to complete a minimum of 18 extra credits either in the same discipline /subject or in different discipline / subject in excess to 90 credits for the Two Year Post Graduate degree programs.

The Add on Proficiency Certification / Diploma so issued to the candidate contains the courses studied and grades earned.

### 9. Assessment and Evaluation

- a) Each course is assessed for a total weight of 100%. Out of the total 100% weight; 50% weight is for Continuous Internal Assessment (CIA or IA) and the remaining 50% for the Semester End Examination (SEE). This applicable for theory, laboratory, workshop, studio and any such courses

b) Out of 50% weight earmarked for Internal Assessment (IA)- 15% for test-1, 15% for test-2 and 20% for Assignments and this is applicable for theory based courses

b)The tests and assignments are conducted as per the semester academic calendar provided by the University

**The details as given in the table**

Note: IA or CIA includes C1 and C2

Component	Description	Conduction	Weight Percentage
<b>C1</b>	<b>Test-1: IA1</b>	<b>8<sup>th</sup> week from the starting date of semester</b>	<b>15</b>
	<b>Test-2: IA2</b>	<b>16<sup>th</sup> week from the starting date of semester</b>	<b>15</b>
<b>C2</b>	<b>1 Assignment</b>	<b>7<sup>th</sup> week</b>	<b>10</b>
	<b>2 Assignment</b>	<b>14<sup>th</sup> week</b>	<b>10</b>
<b>C3</b>	<b>SEE including practical</b>	<b>between 17<sup>th</sup> Week-20<sup>th</sup> Week</b>	<b>50</b>
<b>Results to be Announced</b>			<b>By the end of 21<sup>st</sup> Week</b>

Each test must be conducted for a duration of 60 minutes, setting the test question paper for a maximum of 30 marks. The final examination must be conducted for a duration of 3 hours and the question paper must be set for a maximum of 100 marks.

b.Students are required to complete courses like technical skills, placement related courses, Open electives and any such value addition or specialized courses through online platforms like SWAYAM/NPTEL/Any other reputed online education aggregator. Students are required to choose the courses on the advice of their course coordinator/Director and required to submit the course completion certificate along with percentage of marks/grade scored in the assessment conducted by the online education aggregator. If the online education aggregator has issued a certificate along with the grade or marks scored to students, such courses will be considered for SGPA calculations, in case the aggregator has issued only a certificate and not marks scored, then such courses will be graded through an examination by concerned School, in case, if grading is not possible, students will be given a pass grade and award the credit and the credits will not be considered for SGPA calculations. The Online/MOOCs courses will not have continuous internal assessment component

Such of those students who would like to discontinue with the open elective course that they have already registered for earning required credits can do so, however, they need to complete the required credits by choosing an alternative open elective course.

Setting question paper and evaluation of answer scripts.

For SEE, three sets of question papers shall be set for each theory course out of which two sets will be by the internal examiners and one set will be by an external examiner. In subsequent years by carrying forward the unused question papers, an overall three sets of question papers should be managed and depending on the consumption of question papers either internal or external examiner be called for setting the question paper to maintain an overall tally of 3 papers with the conditioned mentioned earlier. The internal examiner who sets the question paper should have been course tutor

The Chairman of BoE shall get the question papers set by internal and external examiners.

The Board of Examiners shall scrutinize and approve the question papers and scheme of valuation. It is the responsibility of the BoE to see that all questions contained in the question paper are within the prescribed syllabus of the concerned course.

There shall be single valuation for all theory papers by internal examiners. However, there shall be moderation by the external examiner who has the subject background. In case no external examiner with subject background is available, a senior faculty member within the discipline shall be appointed as moderator.

The SEE examination for Practical work / Field work / Project work/Internship will be conducted jointly by internal



and external examiners as detailed below: However, the BoE on its discretion can also permit two internal examiners. If a course is fully of (L=0):T:(P=0) type or a course is partly P type i.e, (L=3): (T=0) (P=1), then the examination for SEE component will be as decided by the BoS concerned.

#### 10. Evaluation of Practical's and Minor Project / Major Project / Dissertation

##### 10.3.1. A practical examination shall be assessed on the basis of:

- a) Knowledge of relevant processes;
- b) Skills and operations involved;
- c) Results / products including calculation and reporting.

##### 10.3.2. In case a course is fully of P type (L=0:T=0:P=4), the performance of a candidate shall be assessed for a maximum of 100 marks as explained below:

- a) Continuous Internal assessment (CIA) = 50 marks
- b) Semester end practical examination (SEE) = 50 marks

The 50 marks for continuous assessment shall further be allocated as under (IA or CIA):

i	Conduction of regular practical throughout the semester	20 marks
ii	Maintenance of lab records /industry reports/SDP reports	15 marks
iii	Laboratory test and viva	15 marks
	Total	50 marks

The 50 marks meant for Semester End Examination, shall be allocated as under:

i	Conduction of semester end practical examination	30 marks
ii	Write up about the experiment / practical conducted	10 marks
iii	Viva Voce	10 marks
	Total	50 marks

10.3.3 The SEE for Practical work will be conducted jointly by internal and external examiners. However, if external examiner does not turn up, then both the examiners will be internal examiners.

10.3.3 In case a course is partly P type i.e, (L=3): (T=0) (P=1), then the examination for SEE component will be as decided by the BoS concerned.

10.3.3 The duration for semester-end practical examination shall be decided by the concerned School Board.

#### 10.3 Evaluation of Minor Project / Major Project / Dissertation:

Right from the initial stage of defining the problem, the candidate has to submit the progress reports periodically and also present his/her progress in the form of seminars in addition to the regular discussion with the supervisor. At the end of the semester, the candidate has to submit final report of the project / dissertation, as the case may be, for final evaluation. The components of evaluation are as follows:

1	First Dissertation presentation describing the problem definition	Should be done a semester before the project semester	Weightage: 0%
2	Dissertation Progress presentation-1	7 <sup>th</sup> week from the start date of project semester	Weightage: 25%
3	Dissertation progress presentation-2	14 <sup>th</sup> Week from the start date of project semester	Weightage -25%
4	Final project Viva and Dissertation Submission	17 <sup>th</sup> -20 <sup>th</sup> Week of project Semester	Weightage: 30% for Dissertation Weightage : 20% for Final Viva Voce

### Provision for Appeal

If a candidate is not satisfied with the evaluation of C1,C2 components, he/she can approach the grievance cell with the written submission together with all facts, the assignments, test papers etc, which were evaluated. He/she can do so before the commencement of semester-end examination. The grievance cell is empowered to revise the marks if the case is genuine and is also empowered to levy penalty as prescribed by the university on the candidate if his/her submission is found to be baseless and unduly motivated. This cell may recommend taking disciplinary /corrective action on an evaluator if he/she is found guilty. The decision taken by the grievance cell is final.

For every program there will be one grievance cell. The composition of the grievance cell is as follows:-

- The Controller of Examination - Ex-officio Chairman / Convener
- One Senior Faculty Member (other than those concerned with the evaluation of the course concerned) drawn from the school / department/discipline and/or from the sister schools / departments/sister disciplines – Member.
- One Senior Faculty Members / Subject Experts drawn from outside the University school / department – Member.

### 12. Eligibility to Appear Semester End Examination (SEE)

**12.1.** Only those students who fulfil a minimum of 75% attendance in aggregate of all the courses including practical courses / field visits etc, as part of the course(s), as provided in the succeeding sections, shall be eligible to appear for SEE examination.

#### 12.2. Requirements to Pass a Course

Students are required to score a total minimum of 40% (Continuous Internal assessment and SEE) in each course offered by the University/ Department for a pass (other than online courses) with a minimum of 20 (40% of 50) marks in final examination

### 13. Requirements to Pass the Semester

To pass the semester, a candidate has to secure minimum of 40% marks in each subject / course of the study prescribed in that semester.

#### 13.1 Provision to Carry Forward the Failed Subjects / Courses:

A student who has failed in a given number of courses in odd and even semesters of first year shall move to third semester of second and final year of the study. However, he / she shall have to clear all courses of all semesters

within the double duration, i. e., within four years of admission of the first semester failing which the student has to re-register to the entire program.

### **13.2 Provision for Supplementary Examination**

In case a candidate fails to secure a minimum of 25% (13 marks) in Semester End Examination (SEE) and a minimum of 40% marks together with IA and SEE to declare pass in the course, such candidate shall seek supplementary examination of only such course(s) wherein his / her performance is declared unsuccessful. The supplementary examinations are conducted after the announcement of even semester examination results. The candidate who is unsuccessful in a given course(s) shall appear for supplementary examination of odd and even semester course(s) to seek for improvement of the performance.

### **13.3. Provision to Withdraw Course:**

A candidate can withdraw any course within ten days from the date of notification of final results. Whenever a candidate withdraws a course, he/she has to register for the same course in case it is hard core course, the same course or an alternate course if it is Soft Core Course or Open Elective Course.

DROPPED course is automatically considered as a course withdrawn.

### **13.4. Re-Registration and Re-Admission:**

a) In case a candidate's class attendance in aggregate of all courses in a semester is less than 75% or as stipulated by the University, such a candidate is considered as dropped the semester and is not allowed to appear for end semester examination (C3) and he / she shall have to seek re-admission to that semester during subsequent semester / year within a stipulated period.

B) In such case where in a candidate drops all the courses in a semester due to personal reasons, it is considered that That he has dropped the semester and he / she shall seek re-admission to such dropped semester.

## **14. Attendance Requirement:**

14.1 All students must attend every lecture, tutorial and practical classes.

**14.2** In case a student is on approved leave of absence (e g:- representing the university in sports, games or athletics, placement activities, NCC, NSS activities and such others) and / or any other such contingencies like medical emergencies, the attendance requirement shall be minimum of 75% of the classes taught.

a) Any student with less than 75% of attendance in aggregate of all the courses including practical courses / field visits etc, during a semester shall not be permitted to appear to the end semester (C4) examination and such student shall seek re-admission as provided in 7.8.4.

b) Teachers offering the courses will place the above details in the School Board meeting during the last week of the semester, before the commencement of C3, and subsequently a notification pertaining to the above will be brought out by the Director of the School before the commencement of C3 examination. A copy of this notification shall also be sent to the office of the Registrar & Controller of Examination.

## **15. Absence during Mid Semester Examination:**

In case a student has been absent from a mid-semester (C1,C2) examination due to the illness or other contingencies he / she may give a request along with necessary supporting documents and certification from the concerned class teacher / authorized personnel to the concerned Head of the School, for make-up examination. The Head of

the School may consider such request depending on the merit of the case and after consultation with course instructor and class teacher, and arrange to conduct a special test for such candidate(s) well in advance before the C3 examination of that respective semester. Under no circumstances C1,C2 test shall be held after C3 examination.

#### 16. Grade Card and Grade Point

**Provisional Grade Card:** The tentative / provisional grade card will be issued by the Controller of Examination at the end of every semester indicating the courses completed successfully. The provisional grade card provides Semester Grade Point Average (SGPA).

**Final Grade Card:** Upon successful completion of M.Sc., Degree a Final Grade card consisting of grades of all courses successfully completed by the candidate will be issued by the Controller of Examination.

**The Grade and the Grade Point:** The Grade and the Grade Point earned by the candidate in the subject will be as given below.

Marks P	Grade G	Grade Point (GP=V x G)	Letter Grade
90 -100	10	v*10	O
80 - 89	9	v*9	A+
70 - 79	8	v*8	A
60 - 69	7	v*7	B+
55 - 59	6	v*6	B
50 - 54	5.5	V*5.5	C +
40 - 49	5	v*5	P
0 - 39	0	v*0	F
ABSENT			AB

O - Outstanding; A+-Excellent; A-Very Good; B+-Good; B-Above Average; C+-Average; C-Satisfactory; F – Unsatisfactory.

Here, P is the percentage of marks ( $P=[C1+C2+C3]$ ) secured by a candidate in a course which is rounded to nearest integer. V is the credit value of course. G is the grade and GP is the grade point.

#### Computation of SGPA and CGPA

The Following procedure to compute the Semester Grade Point Average (SGPA)

The SGPA is the ratio of sum of the product of the number of credits with the grade points scored by a student in all the courses taken by a student and the sum of the number of credits of all the courses undergone by a student in a given

semester, i.e:

$SGPA (S_i) = \sum(C_i \times G_i) / \sum C_i$  where  $C_i$  is the number of credits of the  $i$ th course and  $G_i$  is the grade point scored by the student in the  $i$ th course.

Illustration for Computation of SGPA and CGPA

Illustration No. 1

Course	Credit	Grade Letter	Grade Point	Credit Point (Credit x Grade)
Course 1	4	A+	9	4X9=36
Course 2	4	A	8	4X8=32
Course 3	3	B+	7	3X7=21
Course 4	3	O	10	3X10=30
Course 5	3	P	5	3X5=15
Course 6	3	B	6	3X6=18
Course 7	2	O	10	2X10=20
Course 8	2	A	8	2X8=16
	24			188

Thus,  $SGPA = 188 \div 24 = 7.83$

Illustration No. 2

Course	Credit	Grade letter	Grade Point	Credit Point (Credit x Grade point)
Course 1	4	A	8	4X8=32
Course 2	4	B+	7	4X7=28
Course 3	3	A+	9	3X9=27
Course 4	3	B+	7	3X7=21
Course 5	3	B	6	3X6=18
Course 6	3	P	5	3X5=15
Course 7	2	B+	7	2X7=14
Course 8	2	O	10	2X10=20
	24			175

Thus,  $SGPA = 175 \div 24 = 7.29$

Illustration No.3

Course	Credit	Grade Letter	Grade Point	Credit Point (Credit x Grade point)
Course 1	4	O	10	4 x 10 = 40
Course 2	4	A+	9	4 x 9 = 36
Course 3	3	B+	7	3 x 7 = 21
Course 4	3	B	6	3 x 6 = 18
Course 5	3	A+	9	3 x 9 = 27
Course 6	3	B+	7	3 x 7 = 21
Course 7	2	A+	9	2 x 9 = 18
Course 8	2	A+	9	2 x 9 = 18
	24			199

Thus, SGPA =  $199 \div 24 = 8.29$

Cumulative Grade Point Average (CGPA):

Overall Cumulative Grade Point Average (CGPA) of a candidate after successful completion of the required number of credits (96) for Two year Post Graduate degree program is calculated taking into account all the courses undergone by a student over all the semesters of a program i. e.,  
 $CGPA = \sum(C_i \times S_i) / \sum C_i$

Where  $S_i$  is the SGPA of the  $i$ th semester and  $C_i$  is the total number of credits in that semester.

The SGPA and CGPA shall be rounded off to 2 decimal points and reported in the transcripts.

Illustration: No.1

#### CGPA after Final Semester

Semester (ith)	No. of Credits ( $C_i$ )	SGPA ( $S_i$ )	Credits x SGPA ( $C_i \times S_i$ )
1	24	6.83	24 x 6.83 = 163.92
2	26	7.71	26 x 7.71 = 200.46
3	26	8.68	26 x 8.68 = 225.68
4	14	9.20	14 x 9.20 = 128.8
<b>Cumulative</b>	<b>90</b>		<b>718.86</b>

Thus,  $CGPA = \frac{24 \times 6.83 + 26 \times 7.71 + 26 \times 8.68 + 14 \times 9.20}{90} = \frac{718.86}{90} = 7.99$

#### 16.3.2. CONVERSION OF GRADES INTO PERCENTAGE:

*Conversion formula for the conversion of CGPA into Percentage is:*

*Percentage of marks scored = CGPA Earned x 10*

*Illustration: CGPA Earned 8.10 x 10 = 81.0*

#### 16.3.3. Classification of Results

The final grade point (FGP) to be awarded to the student is based on CGPA secured by the candidate and is given as follows.

CGPA	Grade (Numerical Index)	Letter Grade	Performance	FGP
	G			Qualitative Index
9 >= CGPA 10	10	O	Outstanding	Distinction
8 >= CGPA < 9	9	A+	Excellent	
7 >= CGPA < 8	8	A	Very Good	First Class
6 >= CGPA < 7	7	B+	Good	
5.5 >= CGPA < 6	6	B	Above average	Second Class
> 5 CGPA < 5.5	5.5	C	Average	
> 4 CGPA < 5	5	P	Pass	Satisfactory

Overall percentage=10\*CGPA

#### 17. Challenge Valuation

For all PG courses since it is a double valuation (Internal and External Examiners), candidate

shall not have an option to apply for challenge valuation.

#### Mapping of PEOS with Respect to POS and PSOs

	PO1	P2	PO3	PO4	PO5	PO6	P7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3	PSO4
PEO1	2	1	1	1	1			1			3	1		2	
PEO2	1		2			1			3			3		2	
PEO3			3		2		2			1			3		1
PEO4	1		1		1				3		2	1			3

#### Mapping of Course Outcomes with programme Outcomes

Course Code	PO S/COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PSO1	PSO2	PSO3	PSO4
<b>M18PH1010</b> <b>Mathematical Physics</b>	1	3	3		1	1	1			2	3	2	3			
	2	3	3		1	1	1			2	3	2	3			
	3	3	3		1	1	1			2	3	2	3			
	4	3	3		1	1	1			2	3	2	3			
<b>M18PH1020</b> <b>Classical Mechanics</b>	1	3	3	1	2	2			1			2	3			
	2	3	2	2	2	2			1				3			
	3	3	3	2	3	2	1	1	1			2	3			
	4	3	2	1	1	1			1			2	3			
<b>M18PH1030</b> <b>Electronic devices</b>	1	3	3		1	1	1			2		1				3
	2	3	3	2	1	2						2				3
	3	3	3	2						2		2				3
	4	3	3	2	1	2						2				3

<b>M18PH1040</b> <b>Quantum</b> <b>Mechanics I</b>	1	3	3	3	1	1	1			2		3	3			
	2	3	3	3							2	3	3			
	3	3	3							2		3	3			
	4	3	3								2	3	3			
<b>M18PH1050</b> <b>Material</b> <b>Science</b>	1	3	2	2	2	2	1	1	1			1		3		
	2	3	2	2	2	2	1	1	1			1		3		
	3	3	2	2	2	2	1	1	1			1		3		
	4	3	2	3	3	3	1	1	1			1		3		
<b>M18PH2010</b> <b>Quantum</b> <b>Mechanics</b> <b>II</b>	1	3	3	3	1	1	1				2	3	3			
	2	3	3	3	1	1	1			2		3	3			
	3	3	3		1	1	1				2	3	3			
	4	3	3		1	1	1			2		3	3			
<b>M18PH2020</b> <b>Statistical</b> <b>Mechanics</b>	1	3	3	1	3	2						2	3			
	2	3	3	1	2	1						1	3			
	3	2	3	2	3	3						2	3			
	4	3	2	1	2	2						2	3			
<b>M18PH3010</b> <b>Electrodynam</b> <b>ics</b>	1	3	2		1	1	1			2	1	1	3			
	2	3	2		1	1	1			2	1	1	3			
	3	3	2		1	1	1			2	1	1	3			
	4	3	2		1	1	1			2	1	1	3			
<b>M18PH2040</b> <b>Atomic and</b> <b>Molecular</b> <b>Physics</b>	1	3	2	1	2	2			1			2			3	
	2	3	2	2	2	2			1						3	
	3	3	3	3	3	3	1	1	1			2			3	
	4	2	2	1	1	1			1			2			3	
<b>M18PH2051</b> <b>Electronics-</b> <b>I :Digital</b> <b>electronics</b>	1	3	3	3		1	1	1		2		1				3
	2	3	3	3	1	2						2				3
	3	3	3	3								2				3
	4	3	3	3	1	2						2				3
<b>M18PH2052</b> <b>Condensed</b> <b>Matter</b> <b>Physics - I</b>	1	3	2	1	2	2			1	1		2		3		
	2	3	3	3	3	3	1	1	1	1		2		3		
	3	3	2	2	2	2			1	1				3		
	4	2	2	1	1	1			1	1		2		3		
<b>M20PH2053</b> <b>Material</b> <b>science-I</b>	1	3	3		1	1	1									3
	2	3	2	2	2	2			1							3
	3	3	2	1	2	2	1		1							3
	4	3	2	2	2	2			1							3
<b>M18PH3060</b> <b>Astrophysics</b>	1	3	3	2	1	3	1	1			2	1	3			
	2	3	3	1	3	3						3	3			
	3	3	3	1			1				2		3			
	4	3	3	1				1				1	3			



<b>M20PH3020</b> Nano- science and Nanotechn ology	<b>1</b>	<b>1</b>	<b>2</b>	<b>2</b>	<b>2</b>	<b>2</b>	<b>1</b>	<b>1</b>	<b>1</b>			<b>1</b>		<b>3</b>		
	<b>2</b>	<b>1</b>	<b>2</b>	<b>2</b>	<b>2</b>	<b>2</b>	<b>1</b>	<b>1</b>	<b>1</b>			<b>1</b>		<b>3</b>		
	<b>3</b>	<b>1</b>	<b>2</b>	<b>2</b>	<b>2</b>	<b>2</b>	<b>1</b>	<b>1</b>	<b>1</b>			<b>1</b>		<b>3</b>		
	<b>4</b>	<b>1</b>	<b>2</b>	<b>3</b>	<b>3</b>	<b>3</b>	<b>1</b>	<b>1</b>	<b>1</b>			<b>1</b>		<b>3</b>		
<b>M18PH3030</b> Nuclear and Particle physics	<b>1</b>	<b>3</b>	<b>3</b>	<b>2</b>	<b>3</b>	<b>2</b>	<b>1</b>					<b>2</b>			<b>3</b>	
	<b>2</b>	<b>3</b>	<b>3</b>	<b>2</b>	<b>3</b>	<b>2</b>		<b>1</b>	<b>1</b>			<b>2</b>			<b>3</b>	
	<b>3</b>	<b>3</b>	<b>3</b>	<b>1</b>	<b>3</b>	<b>3</b>						<b>2</b>			<b>3</b>	
	<b>4</b>	<b>3</b>	<b>3</b>	<b>1</b>	<b>2</b>	<b>2</b>		<b>1</b>							<b>3</b>	
<b>M20PH3041</b> Electronics II:	<b>1</b>	<b>2</b>	<b>3</b>							<b>2</b>		<b>1</b>				<b>3</b>
	<b>2</b>	<b>2</b>	<b>3</b>	<b>2</b>	<b>1</b>	<b>2</b>						<b>2</b>				<b>3</b>
	<b>3</b>	<b>2</b>	<b>3</b>	<b>2</b>								<b>2</b>				<b>3</b>
	<b>4</b>	<b>2</b>	<b>3</b>	<b>2</b>	<b>1</b>	<b>2</b>						<b>2</b>				<b>3</b>

**M Sc (Physics) Program**  
**Scheme of Instruction**  
(effective from Academic Year 2018-19)

**SEMESTER-I**

Sl. No.	Course Code	Title of the Course	Type of Course	Credit Pattern & Credit Value				Contact Hours
				L	T	P	Total	
1	M18PH1010	Mathematical Physics	HC	3	1	0	4	5
2	M18PH1020	Classical Mechanics	HC	3	1	0	4	5
3	M18PH1030	Electronic devices	HC	3	1	0	4	5
4	M18PH1040	Quantum Mechanics I	HC	3	1	0	4	5
5	M18PH1050	Material Science (General)	HC	3	1	0	4	5
<b>Practical Courses</b>								
6	M18PH1060	General Physics lab - I	HC	0	0	3	3	5
7	M18PH1070	Electronics lab	HC	0	0	3	3	5
<b>Total Credits</b>				<b>15</b>	<b>5</b>	<b>6</b>	<b>26</b>	<b>35</b>

**SEMESTER-II**

Sl. No.	Course Code	Title of the Course	Type of Course	Credit Pattern & Credit Value				Contact Hours
				L	T	P	Total	
1	M18PH2010	Quantum Mechanics II	HC	3	1	0	4	5
2	M18PH2020	Statistical Mechanics	HC	3	1	0	4	5
3	M18PH2030	Condensed matter physics (General)	HC	3	1	0	4	5
4	M18PH2040	Atomic and Molecular Physics	HC	3	1	0	4	5

5	M18PH2051	Electronics - I : Digital Electronics	SC <sup>#</sup>	2	1	0	3	4
6	M18PH2052	Condensed Matter Physics - I	SC <sup>#</sup>					
7	M18PH2053	Material Science - I	SC <sup>#</sup>					
8	M18PH2060	Music, Dance, sports, Theater, Yoga	RULO	0	0	2	2	3
9	M18PH2070	Skill development Programme	RULO	0	0	2	2	3
<b>Practical Courses</b>								
10	M18PH2080	General Physics lab - II	HC	0	0	3	3	5
11	M18PH2090	Atomic and Molecular Physics	HC	0	0	3	3	5
<b>Total Credits</b>				<b>10</b>	<b>5</b>	<b>8</b>	<b>25</b>	<b>36</b>

**Note: #Soft Core (SC):** Student shall opt for one SC of his/her choice which will be continued in higher semesters also. SEMESTER-III

Sl. No.	Course Code	Title of the Course	Type of Course	Credit Pattern & Credit Value				Contact Hours
				L	T	P	Total	
1	M18PH3010	Electrodynamics	HC	2	1	0	3	4
2	M18PH3020	Nano-science and Nanotechnology	HC	2	1	0	3	4
3	M18PH3030	Nuclear and Particle physics	HC	2	1	0	3	4
4	M18PH3041	Electronics – II: Electronic Communication Systems	SC <sup>#</sup>	3	1	0	4	5
5	M18PH3042	Condensed Matter Physics - II	SC <sup>#</sup>					
6	M18PH3043	Material Science- II	SC <sup>#</sup>					
7	M18PH3051	Electronics – III: Linear integrated circuits	SC <sup>#</sup>	3	1	0	4	5
8	M18PH3052	Condensed Matter Physics - III	SC <sup>#</sup>					
9	M18PH3053	Material Science - III	SC <sup>#</sup>					
10	M18PH3060	Astrophysics	OE <sup>#</sup>	3	1	0	4	5
	Practical Courses							
11	M18PH3070	General Physics lab - III	HC	0	0	3	3	5
12	M18PH3081	Electronics Lab	SC <sup>#</sup>	0	0	3	3	5
13	M18PH3082	Condensed Matter Physics						
14	M18PH3083	Material Science						
Total Credits				15	6	6	27	37

**Note: i) # OE is Open Elective Course** offered for students of other Schools; the students of M Sc – Physics shall take any ONE of the OE course offered by other Schools.

**ii) \*Soft Core (SC):** Students shall have to continue the Soft Core opted during the Second Semester

#### SEMESTER-IV

Sl. No.	Course Code	Title of the Course	Type of Course	Credit Pattern & Credit Value				Contact Hours
				L	T	P	Total	
1	M18PH4010	Major Project	HC	0	0	10	10	20
2	M18PH4020	MOOC/SWAYAM/ Internship	RULO	0	0	4	4	-
3	M18PH4030							-
<b>Total Credits</b>				<b>0</b>	<b>0</b>	<b>18</b>	<b>18</b>	<b>20</b>

<b>Total Credits of I to IV Semesters</b>					
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**Note:** \* The students shall undergo Internship during summer vacation and mid-term vacation soon after Second and Third Semester exams are completed and present the report on the Internship undergone during the Fourth Semester for evaluation. Semester-wise Summary of Credit Distribution

<b>Semesters</b>	<b>No. of Credits</b>	<b>No. of Hours</b>
First Semester	26	35
Second Semester	25	36
Third Semester	27	37
Fourth Semester	18	28
<b>Total Credits</b>	<b>96</b>	<b>136</b>

#### **Distribution of Credits Based on Type of Courses**

<b>Semester</b>	<b>HC</b>	<b>SC</b>	<b>OE</b>	<b>RULO</b>	<b>TOTAL</b>
<b>I</b>	26	-	-	-	26
<b>II</b>	18	3	-	4	25
<b>III</b>	12	11	4	-	27
<b>IV</b>	14	-	-	4	18
<b>Total</b>	<b>70</b>	<b>14</b>	<b>4</b>	<b>8</b>	<b>96</b>

HC=Hard Core; SC=Soft Core; OE=Open Elective;  
RULO=REVA Unique Learning

#### **Distribution of Credits Based on L: T: P**

<b>Semester</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>Total</b>	<b>Total Hours</b>
<b>I</b>	15	5	6	26	35
<b>II</b>	10	5	8	25	36
<b>III</b>	15	6	6	27	37
<b>IV</b>	0	0	18	18	28
<b>Total</b>	<b>40</b>	<b>16</b>	<b>40</b>	<b>96</b>	<b>136</b>

## M. Sc., (PHYSICS) Program

### DETAILED SYLLABUS

#### FIRST SEMESTER

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH1010	MATHEMATICAL PHYSICS	HC	3	1	0	4	5

**Prerequisites:** Basic mathematical knowledge

**Course Objectives:**

To make the students understand the basics of mathematical functions necessary for formulating physical systems and phenomena observed in day to day life.

**Course Outcomes:**

On successful completion of this course, the student shall be able to:

1. Apply the mathematical skills to solve quantitative problems in the study of physics.
2. Solve the problems related to vectors and matrices for various applications.
3. Express a physical law in terms of tensors.
4. The student shall be able to apply coordinate transforms to solve the physical problems expressed in tensors.
5. Apply the suitable mathematical special function to solve the physics problems.

#### Mapping of Course Outcomes with Program Outcomes

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO 11	PS0 1	PS0 2	PS0 3	PS0 4
M18PH1010	CO1	3	3		1	1	1			2	3	2	3			
	CO2	3	3		1	1	1			2	3	2	3			
	CO3	3	3		1	1	1			2	3	2	3			
	CO4	3	3		1	1	1			2	3	2	3			

**Course Content:**

**Unit-1:**

**Vectors:** Review of Vector Algebra, Gradient, Divergence, Curl, Vector Integration, Gauss's Theorem, Stokes's Theorem, Gauss's Law, Laplace equation and Poisson's Equation, Applications of vectors.

**12 hrs**

**Unit - 2:**

**Matrices:** Basic Definitions, Equality, and Rank, Matrix Multiplication, Addition, Inner Product, Matrix Inversion, Orthogonal, Hermitian, and unitary matrices; Diagonalization of Matrices, Matrix representation of linear operators, Eigen values and eigenvectors, characteristic equation of matrix, Cayley-Hamilton theorem, applications of matrices in physics.

**12 hrs**

**Unit - 3:**

**Tensors:** Curvilinear coordinates, Coordinate transformation in linear spaces, Kronecker delta, definition and types of tensors: contravariant and covariant tensors, symmetric and antisymmetric tensors, Tensor algebra: equality, addition and subtraction, tensor multiplication, outer product; contraction of indices, inner product, quotient theorem, metric tensor, Christoffel symbols. Tensors applications in physics. **12hrs**

**Unit - 4:**

**Special functions:** Differential equations, Hermite and Laguerre functions: Partial differential equations, Separation of variables- Helmholtz equation in cylindrical and spherical polar coordinates. Differential equations: Regular and irregular singular points of a second order ordinary differential equation. Series solutions—Frobenius method. Linear independence of solutions-Wronskian. Hermite functions: Generating functions, Recurrence relations, Rodrigues representation, Orthogonality. Laguerre functions: Differential equation-Laguerre polynomials, Generating function, Recurrence relations, Rodrigues representation, Orthogonality, Associated Laguerre functions and its general properties. **12 hrs**

**Reference Books:**

1. Mathematical methods for physicists, Arfken G. B and Weber H.J, 4<sup>th</sup> Edition, Prism Books Pvt Ltd, India (1995).
2. Mathematical Physics, Sathya Prakash, Sultan Chand and Sons, (1985).
3. Mathematical Physics, Chattopadhyaya P.K, Wiley Eastern, (1980).
3. Methods of Mathematical Physics, Bose H.K and Joshi M.C, Tata McGraw Hill, (1984).
1. Vector Analysis, Murray R Spiegel, Schaum's Outline Series, McGraw Hill International Book Company, Singapore (1981).
2. Tensor Analysis — Theory & Applications. Sokolnikoff LS, 211: Edition, John Wiley Sons (1964).
3. Mathematical Methods in the Physical Sciences, Mary L. Boas, 2nd Edition, John Wiley & Sons (1983)

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH1020	CLASSICAL MECHANICS	HC	3	1	0	4	5

**Prerequisites:** Basic knowledge in UG level classical mechanics .

**Course Objectives:**

- To give students a solid foundation in classical mechanics.
- To introduce general methods of studying the dynamics of particle systems.
- To give experience in using mathematical techniques for solving practical problems.
- To lay the foundations for further studies in physics and engineering.

**Course Outcomes:**

On successful completion of this course, the student shall be able to:

1. Apply the Lagrangian methods to analyze and explain the problems in classical Mechanics
2. Apply the concept of Central force problem to analyze and explain the problems Classification of orbits, Kepler's laws in planetary motion.

3. Apply the concept of Hamilton's equations to derive the expression for different principles in classical mechanics
4. Analyse the concept of mechanics of rigid bodies to demonstrate and explain the precession of rotation of earth and Rutherford's scattering in determination of structure of atoms
5. Communicate scientific information of classical mechanics.

#### Mapping of Course Outcomes with Program Outcomes

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO1 0	PO1 1	PS 01	PS0 2	PS0 3	PS0 3
M18PH1020	CO1	3	3	1	2	2			1				2	3		
	CO2	3	2	2	2	2			1					3		
	CO3	3	3	2	3	2	1	1	1				2	3		
	CO4	3	2	1	1	1			1				2	3		

#### Course Content:

##### Unit - 1:

**The Lagrangian method:** Constraints and their classifications. Generalized coordinates. Virtual displacement, D'Alembert's principle and Lagrangian equations of the second kind. Examples of

(I) Single particle in (a) Cartesian coordinates, (b) Spherical polar coordinates and (c) Cylindrical polar coordinates, (II) Atwood's machine. Derivation of Lagrange equation from Hamilton principle. Importance and simple applications of Lagrangian formalism, Symmetry and conservation laws, cyclic coordinates.

**12 hrs**

##### Unit - 2

**Central force problem:** Motion of a particle in a central force field, Conservation of energy and angular momentum, classification of orbits, stability of orbits, Kepler's laws of planetary motion. Scattering in a central potential in Laboratory and centre of mass frames, Impact parameter, Total and differential cross section, Rutherford scattering.

**12 hrs**

##### Unit - 3:

**Hamilton's equations:** Generalized momenta. Hamilton's equations. Examples (i) the simple harmonic oscillator. (ii) Hamiltonian for a free particle in different coordinates. Cyclic coordinates. Physical significance of the Hamiltonian function. Derivation of Hamilton's equations from a variational principle. Generating functions (Four basic types), examples of Canonical transformations, Poisson brackets; properties of Poisson brackets, angular momentum and Poisson bracket relations. Equation of motion in the Poisson bracket notation. The Hamilton-Jacobi equation; the example of the harmonic oscillator treated by the Hamilton-Jacobi method.

**12 hrs**

##### Unit - 4

**Mechanics of rigid bodies:** Degrees of freedom of a free rigid body, Angular momentum and kinetic energy of rigid body. Fixed and moving coordinates, coriolis force, coriolis force acting on falling body. Moment of inertia tensor, principal moments of inertia, products of inertia, the inertia tensor. Euler equations of motion for a rigid body. Torque free motion of a rigid body. Precession of earth's axis of rotation, motion of symmetrical top-rotational motion.

**12 hrs**

**References:**

1. Classical mechanics, H. Goldstein, C. Poole, J. saflco. 3<sup>rd</sup> edition. Pearson Education inc. (2002).
2. Classical mechanics. K. N. Srinivasa Rao, University press (2003).
3. Classical mechanics, N. C. Rana and P.S. Joag Tata McGraw-Hill (1991).
4. Classical dynamics of particles and systems, J. B. Marion, Academic press (1970).
5. Introduction to Classical mechanics. Takwale and Puranik, Tata McGraw-hill (1983)
6. Classical mechanics, L. D. Landau and E. M. Lifshitz, 4<sup>th</sup> edition, Pergamon Press (1985).

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH1030	ELECTRONIC DEVICES	HC	3	1	0	4	5

**Prerequisites: Basic knowledge in electronic devices.**

**Course Objectives:**

- To understand the basic working of Semiconducting devices and Linear Integrated Circuits.
- To give an emphasis to the student to know the various semiconductor devices and its working.
- To give clear understanding of various fabrication techniques of semiconducting devices.
- To introduce the basic building blocks of linear integrated circuits.

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PS11	PS0 1	PS0 2	PS0 3	PSO 4
M18PH1030	CO1	3	3		1	1	1			2		1				3
	CO2	3	3	2	1	2						2				3
	CO3	3	3	2						2		2				3
	CO4	3	3	2	1	2						2				3

**Course Outcomes:**

On successful completion of this course, the student shall be able to

1. Analyse the BJT circuits, operation, and its characteristics.
2. Design a DC bias circuitry of BJT, UJT and SCR.
3. Construct an OPAMP circuit for different applications and develop the prototypes of electronic devices
4. Solve real time examples of BJT, UJT, SCR and OPAMP, and able to describe various opto-electronic devices

**Course Content:****Unit - 1:**

**Transistors:** Transistor configurations and characteristics, Methods of biasing-fixed bias, collector to base bias and voltage divider bias, DC and AC load line, Transistor as an amplifier-Single stage and multistage amplifier, frequency response, Push-pull amplifier, Astable multivibrator using transistors, Voltage regulator using transistors. **12 hrs**

**Unit - 2:**

**Thyristors:** Types of thyristors, working and characteristics of Silicon Controlled Rectifier (SCR), SCR power controller, Characteristics and application of Triac, Working and characteristics of Uni-junction Transistor (UJT), UJT relaxation oscillator. **12 hrs**

**Unit - 3:**

**Operational amplifier:** Block diagram of an operational amplifier, Characteristics of an ideal operational amplifier, Parameters of an op-amp, Operational amplifier as a feedback amplifier: Inverting and Non-inverting amplifiers, Applications of an operational amplifier: Instrumentation amplifier, Square wave and sine wave generator, Active filters- First order Butterworth low pass and high pass filter, phase shift oscillator. **12 hrs**

**Unit - 4:**

**Optoelectronic devices:** Photo resistor (LDR)–dark resistance and material constant, Principle and working of a photodiode, Principle and working of Light emitting diode, factors affecting the efficiency of LED, Phototransistor- structure and working, Semiconductor laser- basic structure and working. **12 hrs**

**Reference Books:**

1. Basic Electronics and Linear Circuits, NN Bhargava, DC Kulashreshtha and SC Gupta, Tata Mc Graw Hill.
2. Electronic Devices and Circuits: An Introduction, Allen Mottershead, Prentice Hall of India.
3. Semiconductor Optoelectronic Devices, Pallab Bhattacharya, Pearson education, Asia.
4. Electronic Principles, A P Malvino, (Sixth Edition, 1999), Tata McGraw Hill, New Delhi.
5. A Text Book of Basic Electronics, RS Sedha, S Chand & Company Ltd.
6. Op-Amps and Linear Integrated Circuits, Remakant A Gayakwad, (Third Edition, 2004), Eastern Economy Edition.
7. Linear Integrated Circuits, D Roy Choudhury and Shail Jain, New Age International Limited.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH1040	QUANTUM MECHANICS – I	HC	3	1	0	4	5

**Prerequisites:** Basic knowledge in wave mechanics and quantum physics



**Course Objectives:**

- To illustrate the inadequacy of classical theories and the need for a quantum theory.
- To explain the basic principles of quantum mechanics.
- To develop solid and systematic problem solving skills.
- To apply quantum mechanics to simple systems occurring in atomic and solid state physics.

**Course Outcomes:**

On successful completion of this course, the student shall be able to:

1. Understand the mathematical representations and analysis used in quantum mechanics.
2. Postulate the basics of quantum mechanics.
3. Apply Schrodinger wave equation for one dimensional problem like, particle in a box, harmonic Oscillator etc.
4. Solve numerical based on angular momentum and spin operators.
5. Analyze the result obtained in Stern-Gerlach experiment.
6. Solve three dimensional problems in quantum mechanics.

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS02	PS03	PS04
M18PH1040	CO1	3	3	3	1	1	1			2			3	3		
	CO2	3	3	3							2		3	3		
	CO3	3	3							2			3	3		
	CO4	3	3								2		3	3		

**Course Content:****Unit - 1**

**Introduction:** Dual nature of matter and waves, Double-slit experiment for photons and electrons as an illustration. Fundamental Postulates of Quantum Mechanics. Review of Linear vector spaces in Dirac Bra-Ket notation. Position and momentum representations. Wavefunctions. Superposition principle. probability densities, probability current. Expectation values. Commutators. Eigenvalues

and eigenvectors of a complete set of mutually commuting operators. Waves, wave packets, phase velocity and group velocity. Canonically conjugate variables, General uncertainty principle. Hamiltonian, Schrodinger's equation. Ehrenfest's Theorem. Continuity equation. **12 hrs**

**Unit - 2**

**Exactly solvable problems in one-dimension:** Bound states, examples of particle in a box, rectangular potential wells, Simple Harmonic Oscillator: wave function and operator approach. particle in a spherically symmetric potential, Rigid rotator, hydrogen atom.

Unbound states, Scattering in one-dimension. Examples of scattering from a one-dimensional rectangular potential well and barrier, Tunneling, Transmission and Reflection co-efficients. Ramsauer - Townsend effect, Alpha decay, cold emission of electron in a metal. **12 hrs**

### Unit - 3

**Angular Momentum and spin:** Angular momentum operators and their Algebra. Eigen functions and Eigen values of  $L^2$  and  $L_z$  using Schrodinger wave mechanics and matrix mechanics. angular momentum operators.

**Uncertainty relations.** Stern-Gerlach experiment and the concept of spin, Pauli-spin matrices. Addition of angular momentum of two or more particles. **12 hrs**

### Unit - 4:

**Exactly solvable problems in three dimensions:** Wave function of a free particle in Cartesian, cylindrical and spherical coordinates. Bound state problems. Examples of a particle confined in a box, cylindrical and spherical well. Simple harmonic oscillator in 3-dimensions. Two-particle bound state problems. Reduction to a one-particle problem. Schrodinger's equation for the hydrogen atom and its solution, properties of its wave functions. **12 hrs**

### Reference Books:

1. E. Merzbacher, Quantum Mechanics. 3rd edition, John Wiley(1994).
2. V.K. Thankappan, Quantum Mechanics, Wiley Eastern (1985).
3. P.M. Mathews and K. Venkatesan, A Textbook of Quantum Mechanics, TMH(1977).
4. R.L.Liboff, Introduction to Quantum Mechanics, Pearson Education(2003).
5. R. Shankar, Principles of Quantum Mechanics, 2nd edition, Plenum US (1994).
6. A Ghatak and S Lokanathan, Quantum Mechanics, Theory and Applications, Macmillan(2004).
7. LI Schiff, Quantum Mechanics, 3rd ed. McGraw-Hill(1968).
8. J.J. Sakurai, Modern Quantum Mechanics, Addison Wesley (1985).
9. B.Bransden, C.Joachain, Quantum Mechanics, 2nd ed, Pearson/Prentice Hall, (2000).
10. J.S.Townsend, A Modern Approach to Quantum Mechanics, 2nd ed, McGraw Hill.
11. C.Cohen-Tannoudji, B.Diu, F.Laloe, Quantum Mechanics (2 vol. set), Wiley Interscience (1996).

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH1050	MATERIAL SCIENCE (GENERAL)	HC	3	1	0	4	5

**Prerequisites: fundamental knowledge in materials.**

### Course Objectives:

- To introduce the basic principles underlying the behavior of materials.
- To provide scientific foundation for understanding the relations among material properties, microstructure, and behavior of materials.
- To make the students familiar with the vocabulary for the description of the empirical facts and theoretical ideas about the various levels of structure from atoms through defects in crystals to larger scale morphology of practical materials.

### Course Outcomes:

On successful completion of this course, the student shall be able to:

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS02	PS03	PS04
M18PH1050	CO1	3	2	2	2	2	1	1	1				1		3	
	CO2	3	2	2	2	2	1	1	1				1		3	
	CO3	3	2	2	2	2	1	1	1				1		3	
	CO4	3	2	3	3	3	1	1	1				1		3	

1. Explain the relationship between a material's microstructure and its properties.
2. Explain how the presence and properties of defects can affect the strength of a material.
3. Distinguish between elastic and plastic behavior of materials.
4. Apply the knowledge of phase diagrams to synthesize different phases of materials.

### Course Content:

#### Unit - 1:

##### Formation and structure of materials

Introduction to material science- engineering materials- structure - property relationship. Review of ionic, covalent and molecular bindings- bond angle, bond length and bond energy. Lattice energy – Jones potential. Closed pack structures- packing efficiency and density of materials.

Crystal morphology - symmetry elements - crystal systems. Point group symmetry- derivation of point groups- elementary ideas on space groups. Principles of X-ray powder diffraction method, interpretation of powder photographs and powder metallurgy.

**12 hrs**

#### Unit - 2:

##### Crystal imperfections and diffusion in solids:

Review of crystalline imperfections- schottky and Frenkel defects- equilibrium concentrations. Line imperfections- edge and screw dislocations-interactions of dislocations. Surface imperfections- grain boundary- tilt and twin boundaries- volume imperfections.

Diffusions in solids - Fick's law of diffusion- Solution to Ficks law - error function. Determination of diffusion co-efficients; diffusion couple. Applications based on second law Atomic model of diffusion- electrical conductivity of ionic crystals.

**12 hrs**

#### Unit - 3:

##### Elastic and plastic behavior of materials

Atomic model of elastic behavior- the modulus as a parameter in design- rubber like elasticity-anelastic behavior – viscosity behavior. Fracture of materials – ductile and brittle fracture – ductile brittle transition- protection against fracture.

Plastic deformation by slip – the shear strength of perfect and real crystals- CRSS- the stress to move a dislocation – work hardening and dynamic recovery. Methods of strengthening crystalline materials against plastic deformation- strain hardening, grain refinement, solid solution strengthening, precipitation strengthening.

**12 hrs**

#### Unit - 4:

##### Phase diagrams and phase transformations

Phase diagrams- the phase rule and it's applications to binary alloy systems- isomorphous, eutectic

and peritectic - the lever rule. Typical phase diagrams-Pb-Sn, Fe-C systems. Heat treatment processes- annealing, hardening and tempering.

Phase transformations- Nucleation and growth- nucleation kinetics – transformations in steel. Solidification and crystallization- recovery, recrystallization and grain growth. Microstructure-single phase materials, phase distribution precipitates and eutectoid decomposition- examples of modifications of microstructure.

**12 hrs**

#### Reference Books:

1. Elements of material science and engineering, **Lawrence H. Van Vlack Addison Wesley** (1975).
2. Material science and engineering, **V. Raghavan**, Prentice Hall (1993)
3. Nature of chemical Bond, **L Pauling**, Oxford and IBH (1960)
4. An introduction to crystallography, **F.C. Phillips**, Longman (1970)
5. Crystallography applied to solid state physics, **Verma and srivastava** New age international (2005)
6. Introduction to solid Solid state physics, **C. kittel**, Wiley Eastern (1993)
7. The structure and properties of Materials vol I- IV- **Rose, Shepard** and wulff (1987)
8. Introduction to solids, **L. V Azaroff, Mc Graw Hill** (1977)
9. Foundation of material science and engineering, William **F. Smith**, Mc Graw Hill international Editions (1988)
10. Solid state Physics Source Book- **Sybil P Parker** (Ed), Mc Graw Hill (1987).
11. Solid state phase transformations, **V. Raghavan**, Prentice hall (1991).

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH1060	GENERAL PHYSICS LAB (PRACTICAL)	HC	0	0	3	3	5

**Prerequisites: fundamental knowledge in handling basic instruments**

#### Course Objectives:

- To make the student familiarize with the basics of experimental physics.
- To enable the student to explore the concepts involved in the thermodynamics and heat.
- To make the student understand the basic concepts in modern optics.
- To allow the student to understand the fundamentals of instruments involved.

#### Course Outcomes:

On successful completion of this course, the student shall be able to:

1. Verify various laws of physics related to optics.
2. Determine the physical parameters through experiments.
3. Analyze the concepts of physics through experiments.

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS0 1	PS0 2	PS0 3	PS0 4
M18PH1050	CO1	3	2	2	2	2	1	1	1				1		3	
	CO2	3	2	2	2	2	1	1	1				1		3	
	CO3	3	2	2	2	2	1	1	1				1		3	
	CO4	3	2	3	3	3	1	1	1				1		3	

#### Course Content:

#### LIST OF EXPERIMENTS:

1. Determination of rigidity modulus and moment of inertia using Torsional pendulum for two different materials.
2. Measurement of resistivity of a semiconductor by Four probe method at different temperature and determination of energy gap.
3. Determination of grating constant and wavelength of LASER light by using grating.
4. Design of CE transistor amplifier.
5. Determination of Stefan's constant and Verification of Stefan's fourth power law by electrical method.
6. Determination of Energy band gap of two different semiconductors.
7. Determination of solar constant.
8. Thermal Conductivity of a rod by Forbe's method.
9. Determination of temperature sensitivity of a thermocouple and its Calibration.
10. Determination of wavelength of sodium light by Michelson's interferometer.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH1070	ELECTRONICS LAB (PRACTICAL)	HC	0	0	3	3	5

**Prerequisites: fundamental knowledge in handling basic electronic instruments**

#### Course Objectives:

- To familiarize the students with the basics of electronics.
- To enable the students to explore the concepts involved in the oscillators.
- To make the students understand the basic concepts in ICs and digital devices.
- To allow the students to understand the fundamentals of multi-vibrators.

#### Course Outcomes:

On successful completion of this course, the student shall be able to

1. Analyze the characteristics of MOSFET and SCR.
2. Verify the outputs of astable, monostable and VCO circuits using ICs.
3. Design and construct the Single Stage BJT and FET Amplifier circuits.
4. Design voltage regulator using Zener diode and regulated power supply using IC.

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PS01	PS02	PS03
M18PH1070	CO1	1	3	3		1	1	1			2	3	2	3
	CO2	2	3	3		1	1	1			2	3	2	3
	CO3	3	3	3		1	1	1			2	3	2	3
	CO4	4	3	3		1	1	1			2	3	2	3

#### Course Content:

#### List of Experiments:

1. Experiment on UJT and its applications.
2. Astable, monostable and bistable multivibrator using IC 555 timer.
3. Voltage controlled oscillator using IC741 and 555.
4. Zener diode characteristics and voltage regulation.
5. Study of FET characteristics and its applications in amplifier.
6. Study of MOSFET characteristics and its applications as amplifier.
7. Characteristics and applications of SCR.
8. Monostable multivibrator using IC 74127.
9. Design of regulated power supply.
10. Solving Boolean expressions.

## SECOND SEMESTER

Course Code	Course Title	Course Type	L	T	P	C	Hrs./Wk.
M18PH2010	QUANTUM MECHANICS –II	HC	3	1	0	4	5

**Prerequisites:** knowledge in quantum mechanics-1

### Course Objective:

To familiarize students with the advanced quantum mechanical concepts for better understanding of behavior of sub-atomic particles.

### Course Outcomes:

On successful completion of this course, the student shall be able to:

1. Solve Schrodinger wave equations for a particle in a 3-D box.
2. Apply Schrodinger wave equation for Hydrogen atom.
3. Understand Symmetry in quantum mechanics.
4. Apply approximation methods for quantum mechanical problems.
5. Understand the concepts of relativistic quantum mechanics.

Course Code	POS/COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS02	PS03	PS04
M18PH2010	CO1	3	3	3	1	1	1				2	3	3			
	CO2	3	3	3	1	1	1			2		3	3			
	CO3	3	3		1	1	1				2	3	3			
	CO4	3	3		1	1	1			2		3	3			

### Course Content:

#### Unit – 1:

**The Schrodinger equation in three dimensions:** separation of Schrodinger equation in Cartesian coordinates. Free particle in 3-d box – Effects of the exclusion principle on non-interacting fermions in a box. central potential and consequences of rotational invariance – separation of variables  $r, \Phi, \Theta$ ; radial equation.

The hydrogen atom – radial equation; energy spectrum; degeneracy of the spectrum; radial wave functions and probability density  $P(r)$  for finding the electron at a distance from the center; evaluation of expectation values of  $r^n$ .

**12 hrs**

#### Unit – 2:

**Symmetry in quantum mechanics:** Spatial transition and conservation of linear momentum. Discrete symmetries: parity and time reversal. Permutation symmetry: symmetric and anti symmetric wave functions for two identical particles, Slater determinant and Pauli exclusion principle. Excited states of

helium atom: ortho and para helium atom.

**Approximation methods – I:** The variational method: variation theorem, application of variation theorem, application of variational approach to ground states of (i) Hydrogen atom and (ii) Helium atom. The WKB method: one dimensional case, approximate solutions turning points and connection formulae, tunneling through a barrier. **12 hrs**

#### **Unit – 3:**

##### **Approximation methods -II:**

Time independent perturbation theory: Perturbation theory for non degenerate states, examples: linear and quadratic stark effects (i) in hydrogen atom, (ii) a particle moving in a 1-d harmonic oscillator. Degenerate perturbation theory, examples: linear stark effect, Normal Zeeman effect.

Time dependent perturbation theory: Time dependent perturbation series. Harmonic perturbation; transition probability, Fermi golden rule, example: sinusoidal perturbation on 1-d simple harmonic oscillator.

Scattering experiments and cross sections: potential scattering, Born approximation, validity of born approximation, Scattering in a central potential, examples: screened coulomb field. **12 hrs**

#### **Unit – 4:**

**Relativistic quantum mechanics:** Klein Gordan equation for a free particle and its drawbacks; probability density. Dirac equation for free particle, properties of Dirac matrices, solutions of free particle Dirac equation- ortho normality and completeness, spin of the Dirac particle, negative energy sea, covariant form of Dirac equation. Velocity operator of a free Dirac particle and Zitterbeugung.

Non relativistic limit of Dirac equation for a free particle moving in a central potential – spin – orbit energy. Dirac particle under the influence of a uniform external magnetic field – magnetic moment for the Dirac particle. **12 hrs**

#### **Reference Books:**

1. Quantum mechanics, **B.H. Bransden and Joachain**, 2<sup>nd</sup> Edition Pearson Education (2004).
2. Introduction to Quantum mechanics, **David J. Griffiths**, 2<sup>nd</sup> Edition, Parson Education (2005).
3. Modern Quantum mechanics, **J.J. Sakurai**, Pearson Education, (2000).
4. Quantum mechanics, V.K Thankappan, 2<sup>nd</sup> Edition 2004.
5. Quantum mechanics, **Stephen Gasiorowicz**, John Wiley (2003).
6. Relativistic Quantum mechanics and Relativistic Quantum fields, J.D. Bjorken and S.D. Drell, Mc. Graw-hill, New York (1968).
7. Quantum mechanics, **L. Sciff Mc. Graw-hill**, (1955).

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH2020	STATISTICAL MECHANICS	HC	3	1	0	4	5

**Prerequisites: fundamental knowledge in thermo dynamics**

#### **Course Objectives:**

- To make students understand the basics of Thermodynamics and Statistical systems.



- To make students understand the various laws of thermodynamic.
- To acquire the knowledge of various statistical distributions.
- To comprehend the concepts of enthalpy, phase transitions and thermodynamic functions.

#### Course Outcomes:

On successful completion of this course, the student shall be able to:

1. Solve day-to-day life selected problems using thermodynamics laws.
2. Analyse various distribution laws.
3. Apply the concepts to test distribution laws.
4. Apply the distribution laws to solve physical problems.

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS02	PS03	PS04
M18PH2020	CO1	3	3	1	3	2						2	3			
	CO2	3	3	1	2	1						1	3			
	CO3	2	3	2	3	3						2	3			
	CO4	3	2	1	2	2						2	3			

#### Course Content:

##### Unit – 1:

**Thermodynamics preliminaries:** A brief overview of thermodynamics, Maxwell's relations, specific heats from thermodynamic relations, the third law of thermodynamics. Applications of thermodynamics: Thermodynamic description of phase transitions, Surface effects in condensation. Phase equilibria; Equilibrium conditions; Classification of phase transitions; phase diagrams; Clausius-Clapeyron equation, applications. Van der Waals equation of state. Irreversible thermodynamics—Onsager's reciprocal relation, thermoelectric phenomenon, Peltier effect, Seebeck effect, Thompson effect, systems far from equilibrium. **12 hrs**

##### Unit – 2:

**Classical statistical mechanics:** The postulate of equal a priori probability; The Liouville theorem; the microcanonical ensemble, canonical ensemble, Grand canonical ensemble, mean value and fluctuations, Entropy and thermodynamic probability, Reduction of Gibbs distribution to Maxwell and Boltzmann distribution, Entropy of an ideal gas; Gibbs paradox; Law of the equipartition theorem; Sackur-Tetrode formula, Molecular partition function, translational and rotational and vibrational partition function and applications to solids. Chemical equilibrium. **12 hrs**

##### Unit – 3:

**Quantum statistical mechanics:** The postulates of quantum statistical mechanics. Symmetry of wave functions. The Liouville theorem in quantum statistical mechanics; condition for statistical equilibrium; Ensembles in quantum mechanics; The quantum distribution functions (BE and FD); the Boltzmann limit of Boson and Fermion gases; the derivation of the corresponding distribution functions. **12 hrs**

##### Unit – 4:

**Applications of quantum statistics:** Equation of state of an ideal Fermi gas (derivation not expected),

application of Fermi-Dirac statistics to the theory of free electrons in metals, degeneracy and magnetic susceptibility. Application of Bose statistics to the photon gas, derivation of Planck's law, comments on the rest mass of photons, Thermodynamics of Black body radiation. Bose-Einstein condensation

**12 hrs**

#### Reference Books

1. Agarwal B.K. and Eisner M., Statistical mechanics, New Age International Publishers, 2000.
2. Roy S.K., Thermal physics and statistical mechanics, New Age International Pub., 2000.
3. Huang K., Statistical mechanics, Wiley-Eastern, 1975.
4. Laud B.B., Fundamentals of statistical mechanics, New Age International Pub., 2000.
5. Schroeder D.V., An introduction to thermal physics, Pearson Education New Delhi, 2008
6. Salinas S.R.A., Introduction to statistical physics, Springer, 2004.

Course Code	Course Title	Course Type		L	T	P	C	Hrs./ Wk.
M18PH2030	<b>CONDENSED MATTER PHYSICS (General)</b>	HC		3	1	0	4	5

**Prerequisites: fundamental knowledge in analysis of materials**

#### Course Objectives:

- To understand the basic knowledge on crystal structures and systems.
- To understand the various process techniques available of X-Ray Crystallography.
- To comprehend the concepts of superconductivity and magnetic properties of solids.

#### Course Outcomes:

On successful completion of this course, the students shall be able to:

1. Formulate basic models of lattice vibrations for describing the physics of crystalline materials.
2. Develop a relation between band structure and the electrical/optical properties of a materials.
3. Analyze the concepts of superconductivity
4. Analyze of the variation of Fermi energy with temperature and impurity concentration in the case of impurity semiconductors.

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS0 2	PS0 3	PSO 4
M18PH2030	CO1	3	2	1	2	2			1	1		2		3		
	CO2	3	3	3	3	3	1	1	1	1		2		3		
	CO3	3	2	2	2	2			1	1				3		
	CO4	2	2	1	1	1			1	1		2		3		

#### Course Content:

##### Unit -1:

**X-ray crystallography:** Crystalline state. Reference axes, equation of a plane, Miller indices. External symmetry of crystals; symmetry operations. Two and three dimensional point groups. Lattices; two dimensional lattices, choice of unit cell. Three-dimensional lattices; crystal systems and Bravais lattices. Screw and glide operations. Space groups; analysis of the space group symbol. Diffraction of Xrays by crystals: Laue equations. Reciprocal lattice. Bragg equations. Equivalence of Laue and Bragg equations.

**Unit - 2:**

**Experimental techniques:** Brief introduction to Laue, Oscillation, Weissenberg, Powder and Counter methods. Using synchrotron radiation for structure studies.

**Electron and neutron diffraction:** Basic principles. Differences between them and X-ray diffraction. Applications (qualitative).

**Crystal growth:** Crystal growth from melt and zone refining techniques. Liquid crystals: Morphology. The smectic (A-H), nematic and cholesteric phases. Birefringence, texture and X-ray studies in these phases. Orientational order and its determination in the case of nematic liquid crystals. **12 hrs**

**Unit -3:**

**Magnetic properties of solids:** Diamagnetism and its origin. Expression for diamagnetic susceptibility, Paramagnetism. Quantum theory of Paramagnetism. Brillouin function. Ferromagnetism. Curie-Weiss law. Spontaneous magnetization and its variation with temperature. Ferromagnetic domains. Antiferromagnetism. Two sub-lattice model. Susceptibility below and above Neel's temperature.

**Superconductivity:** Experimental facts. Type I and type II semiconductors. Phenomenological theory. London equations. Meissner effect. High frequency behavior. Thermodynamics of superconductors. Entropy and Specific heat in the superconducting state. Qualitative ideas of the theory of superconductivity. **12 hrs**

**Unit - 4:**

**Semiconductors:** Intrinsic Semiconductors. Crystal structure and bonding. Expressions for carrier concentrations. Fermi energy, electrical conductivity and energy gap in the case of intrinsic semiconductors. Extrinsic Semiconductors; impurity states and ionization energy of donors. Carrier concentrations and their temperature variation. Qualitative explanation of the variation of Fermi energy with temperature and impurity concentration in the case of impurity semiconductors.

**12 hrs****Reference Books:**

1. Stout G.H. and Jensen L.H., X-ray structure determination, MacMillan, USA, 1989.
2. Ladd M.F.C. and Palmer R.A., Structure determination by X-ray crystallography, Plenum Press, USA, 2003.
3. Buerger M.J., Elementary crystallography, Academic Press, London.
4. Dekker A.J., Solid state physics, Prentice Hall, 1985.
5. Kittel C., Introduction to solid state physics, 7th Edn., John Wiley, New York, 1996.
6. McKelvey J.P., Solid state and semiconductor physics, 2nd Edn., Harper and Row, USA, 1966.
7. Streetman B.G., Solid state electronic devices, 2nd Edn., Prentice-Hall of India, New Delhi, 1983.
8. De Gennes P.G. and Prost J., The physics of liquid crystals, 2nd Edn., Clarendon Press, Oxford, 1998.
9. Wahab M.A., Solid state physics, Narosa Publishing House, New Delhi, 1999.
10. Azaroff L.V., Introduction to solids, McGraw-Hill Inc, USA, 1960.
11. Pillai S.O., Solid state physics, New Age International Publications, 2002.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./Wk.
M18PH2040	ATOMIC AND MOLECULAR PHYSICS	HC	3	1	0	4	5

**Prerequisites:** fundamental knowledge in Atomic Physics

**Course Objectives:**

- To develop a basic understanding of physics of atoms and molecules: definitions, units, laws and rules.
- to gain an ability of basic problems analyzing and solving in physics of atoms and molecules
- to realize a role and practical application of physics of atoms and molecules in the modern world

**Course Outcomes:**

On successful completion of this course, the student shall be able to:

1. Analyze the concepts of atomic models, spectra of one and two valance electron atoms.
2. Analyze the behaviour of atoms in external applied electric and magnetic field.
3. Differentiate rotational, vibrational, electronic and Raman spectra of molecules.
4. Describe electron spin and their spectroscopic applications.
5. Formulate and solve the problems related to different spectroscopic systems.

Course Code	POS/COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS02	PS03	PS04
M18PH2040	CO1	1	3	2	1	2	2			1			2			3
	CO2	2	3	2	2	2	2			1						3
	CO3	3	3	3	3	3	3	1	1	1			2			3
	CO4	4	2	2	1	1	1			1			2			3

**Course Content:**

**Unit – 1:**

**Atomic Physics:** Brief review of early atomic models of Bohr and Sommerfeld. One electron atom: Quantum states, Atomic orbitals, spectrum of hydrogen, Rydberg Atoms (brief treatment), Relativistic corrections to spectra of alkali atoms: Spin-orbit interaction and fine structure in alkali spectra. Lamb shift in hydrogen (qualitative Discussion only). **12 hrs**

**Unit – 2:**

Two electron atom: Ortho and Para states and role of Pauli principle, level schemes of two electron atoms. Perturbations in the spectra of one and two electron atoms: Zeeman effect, Paschen- Back effect, Stark effect in hydrogen spectra. Hyperfine interactions and 21cm line of hydrogen. Many electron atoms: Central field approximation. LS and JJ coupling schemes, Multiplet splitting and Lande interval rule. **12 hrs**

**Unit – 3:**

**Molecular Physics A:** Brief treatment of chemical bonds: covalent, ionic, Vanderwaal's interactions. The Born-Oppenheimer approximation (qualitative treatment), diatomic molecule as a rigid rotator, rotational spectra of rigid and non-rigid rotator, intensities of rotational lines. Microwave spectroscopy-principle and technique Types of rotors: Eigen values of Linear, Symmetric top, Asymmetric top and Spherical top molecules. Raman spectroscopy: Theory of Raman effect, experimental techniques, rotational Raman spectra of diatomic and linear polyatomic molecules.

**12 hrs****Unit – 4:**

**Molecular Physics B:** Diatomic molecule as a simple harmonic oscillator, anharmonicity, Morse potential curves, vibrating rotator: energy levels and vibration spectra, PQR branches in rovibronic spectra, experimental technique and IR spectrometer. Comparison of vibration and Raman spectra.

**Electronic spectra of diatomic molecules:** Vibrational structure, rotational structure in electronic spectra, intensity of vibrational lines in electronic spectra, Frank-Condon principle, dissociation and pre-dissociation, fluorescence and phosphorescence.

**12hrs****Reference Books:**

1. Introduction to Atomic spectra- H.E.White.
2. Fundamentals of molecular spectroscopy, C B Banwell.
3. Spectroscopy Vol I, II & III, Walkere and Straughen.
4. Physics of atoms and molecules, Bransden and Joachain, (2nd Edition) Pearson Education, 2004.
5. Fundamentals of Molecular Spectroscopy, Banwell and Mccash, Tata McGraw Hill, 1998.
6. Modern Spectroscopy, J.M. Hollas, John Wiley, 1998.
7. Molecular Quantum Mechanics, P.W. Atkins and R.S. Friedman. Third Edition, Oxford Press(Indian Edition), 2004.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH2051	ELECTRONICS-I (DIGITAL ELECTRONICS)	SC	2	1	0	3	4

**Prerequisites: fundamental knowledge in general electronics****Course Objectives:**

- To understand number systems and codes and their application to digital circuits.
- To gain knowledge of Boolean algebra, Karnaugh maps and its application to the design and characterization of digital circuits.
- To know the mathematical characteristics of logic gates.
- To design and analyze a given combinational or sequential circuit using Boolean Algebra as a tool to simplify and design logic circuits.
- To understand the logic design of programmable devices, including PLDs
- To design various synchronous & Asynchronous counters and Universal Shift Registers.

**Course Outcomes:**

On successful completion of this course, the student shall be able to:

1. Translate from one number system to its equivalent other number system and perform its arithmetic operations.
2. Explain TTL and CMOS construction, working, characteristics and applications.
3. Explain the use of implementation of logic circuits.
4. Draw the logic circuits by simplifying Boolean expressions using theorems, laws and K-map.
5. Analyse the operation of Combinational and Sequential logic circuits.
6. Explain DAC and ADC, types, specifications and applications.

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS0 2	PS0 3	PSO 4
M18PH2051	CO1	3	3	3		1	1	1		2		1				3
	CO2	3	3	3	1	2						2				3
	CO3	3	3	3								2				3
	CO4	3	3	3	1	2						2				3

### Course Content:

#### Unit - 1:

**Number Systems and Logic families:** Decimal, Binary, octal, and hexa-decimal number systems, binary arithmetic. Number base conversion, Complements Codes: Binary code, excess-3 code, gray code, error detection and correction codes.

Positive logic and Negative Logic, AND, OR, NOT, NAND, NOR, X-OR GATE, INHIBIT CIRCUIT, Significance and type like TTL, CMOS, interface with different logic families, application relevant information, electrical characteristics. **12 hrs**

#### Unit - 2:

**Boolean operations and expressions:** Introduction, Logic Operators, Postulates and theorems, properties –Product of Sums and Sum of Products– Karnaugh Map method – Two, three, four, five variable K-maps, Converting Boolean expressions to Logic and Vice versa, NAND and NOR implementation – Don't-Care conditions – The tabulation method. **12 hrs**

#### Unit - 3:

**Combinational and Sequential circuits:** Half and full Adder – Half and full Subtractor – Binary parallel adder – BCD Adder, Decimal adder – Magnitude comparator – Encoders & Decoders – Multiplexers– De-multiplexer Latches, Flip-flops, SR, JK- Flip-flop, JK Master-Slave, D, T flip-flops, counters, synchronous and asynchronous counters, ripple counters, registers, shift registers, timing sequences. **12 hrs**

#### Unit - 4:

**A/D and D/A conversion circuits:** Introduction, Digital to Analog Converters D/A converter Specifications, Types of D/A converters, Mode of Operation, BCD Input D/A converter, Integrated Circuit D/A Converters, D/A converter Applications, A/D converters, A/D Converter Specifications, A/D Converter Technology, Types of A/D converters, Integrated Circuit A/D Converters, A/D converter Applications

**Basics of microprocessor and microcontroller:** Architecture of 8085, Architecture of 8051. **12 hrs**

**Reference Books:**

1. John F. Wakerly, "Digital Design" 4th edition, Pearson/PHI, 2008.
2. John, M Yarbrough, "Digital Logic application and design", Thomson Learning, 2006.
3. Charles H, Roth, "Fundamentals of Logic Design", Thomson Learning, 2013.
4. Donald P, Leach and Albert Paul Malvino, " Digital Principles and Applications", 6<sup>th</sup> edition, TMH, 2006.
5. Thomas L. Floyd, " Digital Fundamentals", 10<sup>th</sup> Edition, Pearson Education Inc, 2011
6. Donald D, Givone, "Digital Principles and Design", TMH, 2003.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH2052	CONDENSED MATTER PHYSICS - I	SC	2	1	0	3	4

**Prerequisites: subject knowledge in CONDENSED MATTER PHYSICS**

**Course Objectives:** To familiarized students with:

- Defects, types of defects and how defects are formed.
- Luminescence and its mechanism.
- Lattice formation and deformation, energy band structure and classification of materials based on energy gap.
- Preparation of thin films and study of their structural characteristics.

**Course Outcomes:**

On successful completion of this course, the student shall be able to:

1. Analyze defects in crystals.
2. Illustrate Photoconductivity, Luminescence- fluorescence, Phosphorescence, Thermoluminescence, Photoluminescence, Electroluminescence, Mechanisms.
3. Analyze Short range order- long range order.
4. Design semiconductor devices.

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS0 2	PS0 3	PSO 4
M18PH2052	CO1	3	2	1	2	2			1	1		2		3		
	CO2	3	3	3	3	3	1	1	1	1		2		3		
	CO3	3	2	2	2	2			1	1				3		
	CO4	2	2	1	1	1			1	1		2		3		

## Course Content:

### Unit -1:

**Disordered systems:** Point defects-shallow impurity states in semiconductors-Localized lattice vibrational states in solids-Vacancies, interstitials in ionic crystals- Color centers in ionic crystals- Types of Color centers- methods of production-mechanism - Characteristic absorption bands, Properties of Color centers- Models and Applications.

Photoconductivity, Luminescence- fluorescence, Phosphorescence- Thermoluminescence, Photoluminescence, Electroluminescence; Mechanisms. Imperfections in crystals, Mechanism of plastic deformation in solids, Stress and strain fields of screw and edge dislocations, Elastic energy of dislocations.

**12 hrs**

### Unit – 2:

**Disorder in Condensed Matter :** Introduction- Short range order- Long range order- Ordered lattice- Disordered lattice- Compositional. disorder- Topological disorder-Magnetic disorder-Localized states- Anderson Model- Density of states. Concept of glass- Glass transition- Atomic correlation function and structural description of glasses and liquids. Amorphous semiconductors: Classification, band structure- electronic conduction- Optical absorption-Switching. Transport in disordered lattices- Transport in extended states, Fixed range and variable range hopping- conductivity in impurity bands and in amorphous semiconductors-Applications.

**12 hrs**

### Unit – 3:

**Semiconductors:** Structure of typical semiconductors- Fermi energy expression in Intrinsic and extrinsic semiconductors- its variation with temperature and impurity concentration- Law of mass action- Charge neutrality equation- mobility- diffusion- generation- recombination of Carriers in Semiconductors- Conductivity equation- carrier Life time- Haynes-Shockley experiment- Hall effect in semiconductors- Determination of Hall coefficient in intrinsic, n-type and p-type semiconductors.

**12 hrs**

### Unit – 4:

**Films and Surfaces:** Thin films Methods of preparation: Thermal evaporation- sputtering- DC, AC, diode, triode, magnetron, ion beam sputtering, Laser and electron beam evaporation technique. Chemical vapor deposition. Characterization of thin films- film thickness: optical methods- interferometry- Fizeau fringes- FECO Method. Mechanical techniques- Stylus method-weight measurement and crystal oscillators. Structural characterization Scanning electron microscopy, Transmission Electron microscopy and Atomic Force Microscopy. Mechanical properties- Internal stress and strain analysis. Electrical properties of thin films- Measurement of resistivity by four probe method, thin film resistors (Conduction in metal and non metallic films) Magnetic properties- film size effect on MS- films for memory devices.

**12 hrs**

## Reference Books:

1. Solid State Physics, A. J. Dekker, McMillan India Ltd, 2003.
2. Luminescence of Solids, D. R. Vij, Plenum Press, 1998.



3. Elementary dislocation theory, J. Weertman and J.R. Weertman, New York ; Macmillan 1964
4. Crystallography Applied to Solid State Physics, Verma and Srivastava, 2<sup>nd</sup> Edition. New age International publishers, 2001.
5. Introduction to Solid State Physics, C. Kittel, 7<sup>th</sup> Edition, John Wiley and Sons 1996.
6. Thin Film phenomenon, K.L. Chopra, McGraw Hill Book Company, 1969.
7. Introduction to solid state theory, Otfried Madelung, Springer series. 1996.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./Wk.
M18PH2053	MATERIAL SCIENCE –I	SC	2	1	0	3	4

### Prerequisites: subject knowledge in MATERIAL SCIENCE

#### Course Objectives:

1. To focus on the structural, electronic and magnetic properties of metals, alloys, super conductors, semi-conductors and dielectric materials.
2. To discuss and understand various applications of the above materials in different fields.

#### Course Outcomes:

On successful completion of this course, the student shall be able to:

1. Explain the Structure, types of metals and alloys.
2. Analyze the effect of doping and mobility of charge carriers in semiconductors.
3. Analyze the concept of dielectrics and magnetic materials phenomena.
4. Analyze the Properties of Magnetic Materials based on resonance and spin configurations.

Course Code	POS/COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS02	PS03	PS04
M18PH2053	CO1	3	2	2	2	2	1	1	1			1		3		
	CO2	3	2	2	2	2	1	1	1			1		3		
	CO3	3	2	2	2	2	1	1	1			1		3		
	CO4	3	2	3	3	3	1	1	1			1		3		

#### Course Content:

##### Unit - 1:

**Metals:** Review of free electron theory and Fermi distribution function, Structure and types of metals, Electronic properties of metals- electrical and thermal conductivity, Widemann- Franz law, temperature and impurity effects. Heat capacity of metals- debye's model of specific heat-contribution of free electrons to heat capacity- dispersion relation- acoustic and optical modes-thermal expansion- an harmonic interactions, Galvanomagnetic effects in metals.

**Alloys:** Solid solutions - substitutional and interstitial. Hume Rothery rules. Super lattice- long range order theory. Diffusion in alloys- Darkens equation. Some special alloys-ferrous and nonferrous, super alloys.

**12 hrs**

## Unit - 2:

**Semiconductors:** Review of band theory of solids, direct and indirect band gaps, charge carrier in intrinsic semiconductor. Extrinsic semiconductor- effect of doping and mobility of charge carriers. Methods of doping- alloying, diffusion and ion implantation. Preparation of semiconductor single crystals.

**Superconductivity:** Superconducting tunneling phenomena. AC and DC Josephson effect. Applications- Superconducting magnets, super density switches.

**SQUID.** HTS superconductors - materials preparation and structure.

**12 hrs**

## Unit - 3:

### Dielectrics and Ferroelectrics

**Dielectrics:** Review of dielectric polarization- internal field and macroscopic field. The Complex dielectric constant-dielectric losses and relaxation time-Debye equations- Theory of electronic polarization and optical absorption. Dielectric function» LST Relationship, dielectric breakdown-general applications of dielectric materials. Ferroelectrics Piezoelectric, pyroelectric and ferroelectric materials- transducer and detector applications, Classification of ferroelectrics.

Ferro electricity in KDP and barium titanate- order—disorder and displacement theories. Thermodynamics of ferroelectric phase transitions.

**12 hrs**

## Unit - 4:

### Magnetic Materials and Magnetic Resonance:

**Magnetic Materials:** Review of dia, para and ferro- magnetic materials, Spontaneous Magnetization— temperature dependence- gyromagnetic experiments. Origin of Ferromagnetic domains- anisotropy of magnetostriction and Bloch wall energies. Antiferromagnetic and ferrimagnetism- Sublattice model~ Neel's theory. Neutron Diffraction in magnetic structure analysis. Hard and soft magnetic materials- areas of Their application.

**Magnetic Resonance:** Elements of theory of nuclear magnetic resonance (NMR)-rate of energy absorption- spin lattice and spin-spin relaxation- Bloch equations, Principles of ESR, NMR and Mossbauer techniques, typical areas of application.

**12 hrs**

### Reference Books:

1. Introduction to Properties of Materials — D. Rosenthal and R M Asirnov, East West (1974).
2. Elements of Materials Science and Engineering- L H Van Vlack, Addison Wesley (1975).
3. Introduction to solid state Physics, C. Kittel, Wiley Eastern (1993).
4. Solid State Physics, A. J. Dekker, Mc Milan India (2005).
5. Introduction to solids, L V Azaroff, Mc Graw Hill (1977).
6. Electronic Materials, S. Muraka, Academic Press (1989).
7. Superconductivity and Superconducting Materials- A. V. Narlikar and S. N. Ekbote, South Asian Publications (1983).
8. Semiconductor Physics- P S Kireev, Mir Publishers (1975).
9. Solid State and semiconductor Physics, John Mckelvey, Harper and Low (1969).
10. Modern Magnetism- L F Bates, Cambridge University Press(1963).

11. Electronic Properties of Materials Ver, Hummel, Springer lag (1985).
12. Physics of dielectric Materials- 13 Tareev, Mir Publishers (1979).
13. Magnetic Resonance- C P Slichter , Harper and Row (1985).
14. NQR Spectroscopy, SSP Suppl. I T P Das and E. L. Hahn, Academic Press (1957).
15. Mossbauer Effect and its Applications, V G Bhide, Tata McGraw Hill (1973).

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH2060	MUSIC / DANCE / SPORTS / THEATER / YOGA	RULO	0	0	2	2	3

**Note:** Music, Dance, and Theater courses are offered by the School of Performing Arts, whereas the Sports and Yoga courses are offered by the Department of Physical Education. The students have to choose any **ONE** of these courses.

#### A. YOGA FOR HEALTH

##### Course Objectives:

Following are the Course Objectives.

- To prepare the students for the integration of their physical, mental and spiritual faculties;
- To enable the students to maintain good health;
- To practice mental hygiene and to attain higher level of consciousness;
- To possess emotional stability, self control and concentration; and
- To inculcate among students self discipline, moral and ethical values.

##### Course Outcomes:

On completion of the course learners will be able to:

- 1.Practice yoga for strength, flexibility, and relaxation.
- 2.Learn techniques for increasing concentration and decreasing anxiety
- 3.Become self disciplined and self-controlled
- 4.Improve physical fitness and perform better in studies
- 5.Gain self confidence to face the challenges in the society with commitment to serve the society

##### Course Contents

###### Unit-I:

**Yoga:** Introduction, Tips from Sage Patanjali's Yoga Sutras

**Surya Namaskara:-** 10 counts,12 counts,16 counts

###### Unit-II:

**Asanas: Sitting-** Vajrasana, Dandasana, Padmasana, Matsyasana, Ardha Matsyendrasana, Suptavajrasana, Paschimottasana, Bakasana, Simhasana, Shirasasana.

**Asanas: Standing-** Tadasana, Trikonasana, Parshwa konasana, Veerabhadrasana, Parivrtta trikonasana.

**Unit-III:**

**Asanas: Prone Position-** Bhujangasana, Dhanurasana, Shalabhasana.

**Asanas: Supine Position-** Sarvangasana, Sethubandha sarvangasana, Halasana, Karnapeedhasana.

**Mudras-** Dhyana mudra, Chinmaya mudra, Namaste mudra, Nasika mudra

**Unit-IV:**

**Pranayams:-** Ujjayi, Nadi Shodhana, Anuloma – Viloma, Basthrika, Bhramari, Sheethali

**Dhyana & its types**

Competition format, Rules and their interpretations

**B. SPORTS (VOLLEYBALL)**

**Course Objectives:**

1. To learn the rules, fundamental skills, and strategies of volleyball.
2. To develop skills in passing, setting, serving, spiking, and blocking.
3. To learn basic offensive and defensive patterns of play.
4. To develop a positive attitude towards volleyball as a lifetime sport and to improve physical fitness through participation in volleyball.

**Course Outcomes:**

On completion of the course learners will be able to:

1. Learn basic skills and knowledge associated with volleyball.
2. Apply these skills while playing volleyball and exhibit improved performance
3. Improve physical fitness and practice positive personal and lifestyle.
4. Gain an understanding of the value of sports in attaining wellness, maintaining good health and developing spirit of teamwork.

**Course Contents:**

**Unit-I**

- **Introduction about Volleyball**
- Players Stance, Receiving and passing
- The Volley (Overhead pass), The Dig (Underhand pass), Service Reception

**Unit-II**

- Service- Under Arm Service, Tennis Service, Side Arm Spin Service, Round Arm Service, High spin service, Asian serve / American serve (floating)
- Setting the ball- Set for attack, Back set, Jump set

### **Unit-III**

- Smash/Spike- Straight smash, Body turn smash, Wrist outward smash, Wrist inward smash
- Block- Single block, Double block, Three-man block
- Rolls- Overhead pass & back rolling, One hand underhand pass with side rolling, Forward dive

### **Unit-IV**

- Attack Combination, Defense Systems, Libero play
- Court marking, Rules and their interpretations and Duties of officials

## **C. SPORTS (BASKETBALL)**

### **Course Objectives:**

1. To learn the rules, fundamental skills, and strategies of Basketball
2. To develop technical skills in passing, in ball handling, individual offense, individual defense, rebounding, screen, team offense, team defense and fast break.
3. To learn basic offensive and defensive strategies of play.
4. To develop a positive attitude towards Basketball as a lifetime sport and to improve physical fitness through participation in Basketball.
5. To develop positive understanding and appreciation of the basketball game.

### **Course Outcomes:**

On completion of the course learners will be able to:

1. Learn basic skills and knowledge associated with basketball.
2. Apply these skills while playing basketball and exhibit improved performance
3. Improve physical fitness and practice positive personal and lifestyle.
4. Gain an understanding of the value of sports in attaining wellness, maintaining good health and developing spirit of teamwork.

### **Course Contents:**

#### **Unit-I**

- Basketball: Introduction
- Grip; Player stance- Triple threat stance and Ball handling exercises
- Passing (Two hand/one hand)- Chest pass, Bounce Pass, Over head pass, Underhand pass, Hook Pass, Behind the back pass, Baseball pass, Side arm pass and passing in running.
- Receiving-Two Hand receiving, One hand receiving, Receiving in stationary position, Receiving while jumping, Receiving while running.

#### **Unit-II**

- Dribbling- How to start dribble, How to stop dribble, High / Low dribble with variations
- Shooting- Layup shot and its variations, One hand set shot, One hand jump shot, Free throw, Hook shot, Tip-in shot.

- Stopping- Stride/Scoot, Pivoting and Faking /Feinting footwork.

#### **Unit-III**

- Rebounding- Defensive rebound, Offensive rebound, Box out, Rebound Organization.
- Individual Defensive- Guarding the man with the ball and without the ball.
- Offensive drills, Fast break drills, Team Defense/Offense, Team Tactics

#### **Unit-IV**

- Court marking, Rules and their interpretations

### **D. SPORTS (FOOTBALL)**

#### **Course Objectives:**

1. To learn the rules, fundamental skills, and strategies of football.
2. To develop skills in passing, receiving, controlling the ball, dribbling, shielding, shooting, tackling, beating a defender and heading in football.
3. To learn basic offensive and defensive patterns of play
4. To use different parts of the body in utilizing the above skills while playing football
5. To develop a positive attitude towards football as a lifetime sport and to improve physical fitness through participation in football.

#### **Course Outcomes:**

On completion of the course learners will be able to:

1. Learn basic skills and knowledge associated with football.
2. Apply these skills while playing football and exhibit improved performance
3. Use the knowledge and understanding to perform, refine and adapt the above skills and related skills with precision, accuracy, fluency and clarity in any situation.
4. Improve physical fitness and practice positive personal and lifestyle.
5. Gain an understanding of the value of sports in attaining wellness, maintaining good health and developing spirit of teamwork.

#### **Course Content:**

##### **Unit-I**

##### **Football: Introduction**

Kicks- Inside kick, Instep kick, Outer instep kick, Lofted kick, Chipping, Volley, Half Volley

Trapping- Trapping rolling the ball, Trapping bouncing ball with sole

##### **Unit-II**

- Dribbling- With instep and outer instep of the foot.
- Heading- From standing, running and jumping.
- Feinting- With the lower limb and upper part of the body.

### **Unit-III**

- Tackling- Simple tackling, Slide tackling.
- Throw-in- Standing and Sliding
- Goal Keeping- Collection of balls, Ball clearance, throwing and deflecting.

### **Unit-IV**

- Ground marking, Rules and their interpretations

## **E. SPORTS (TRACK AND FIELD )**

### **Course Objectives:**

1. To teach students the skilled techniques in sprints, relay running, hurdles, long jump, high jump, and shot put and practice them.
2. To develop competence among students in demonstrating all the techniques covered in the course.
3. To make students understand some of the scientific and empirical principles and their rationale underlying the development of skilled performance.
4. To inculcate among students the habit of team work and cooperative learning and develop competence in detecting / correcting technique errors.
5. To develop a positive attitude towards sports in general and athletics in particular and to improve physical fitness through participation in various athletic games / sports activities.

### **Course Outcomes:**

On completion of the course learners will be able to:

1. Display competencies in executing basic techniques and skills associated with select track and field events.
2. Develop basic skills and techniques to improve one's running posture and take-off position for different jumps.
3. Learn regular practice of select track and field events and improve physical fitness
4. Appreciate track and field events by applying sports science knowledge to explain the execution of the events.

### **Course Content:**

#### **Unit-I**

Athletics: Introduction

Track Events - Steeple Chase, Race Walking, Middle and Long distance races

Race walking - Technique, Faults and Officiating.

Middle and Long distance races – Technique and Training

#### **Unit-II**

Jumping Events - High Jump and Triple Jump: Basic Skills and techniques

High Jump - Straddle Roll & Flop Technique, Approach, Take-off, Technique in the air, Clearance over the bar & Landing

Triple Jump – Hop, Step and Jump Technique, Approach, Take-off & Landing

### **Unit-III**

Throwing Events - Discus Throw and Hammer Throw: Basic Skills and techniques  
Discus Throw - Standing and Rotatory techniques, Grip, Stance, Rotation Technique, Power stance, Release and Reverse (Follow through)  
Hammer Throw - Grip, Swings, Rotation foot work, Release and Follow through

### **Unit-IV**

Rules, Officiating and Marking - Ground / Sector Marking, Interpretation of Rules.

### **Reference Books**

#### **(Athletics Part-I and Athletics Part-II)**

1. Arthur E. Ellison (ed) (1994). Athletic Training and Sports Medicine.
2. Ballisteros, J.M. (1998). Hurdles Basic Coaching Manual, IAAF.
3. Bosen K.O. (1993). Teaching Athletics Skills and Technique.
4. Bosen K.O. (1990). Study Material on Hurdles for the Regular Course Students.
5. Doherty K. (1995). Track and Field Omni book.
6. Martin, David E. Peter N. Coe (1991). Training Distance Runner.
7. Howard S. (1981). Science of Track and Field Athletics.
8. Briggs Graeme (1987). "Track and field coaching Manual", Australian Track and Field Coaches Association. Rothmans Foundation National Sports Division.
9. Carr, Gerry (1999). "Fundamentals of Track and Field. Track Athletics 1 Title G.V. 1060 5.e. 368.
10. I.A.A.F. Level-II (2001). Text Book on Jumping Event.
11. Jarver, Jesse (1987). "The Jumps", Track and Field Coaching Manual Australia.

### **F. DRAMATICS**

**Pre-requisites:** Students with background in Theatre Arts/ Keen interest in Dramatics.

#### **Course Objectives:**

- To imbibe the acting skills.
- To understand the broader applications of theatre studies in allied arts forms.
- To be able to use body language for better communication.
- Students shall also be able to understand voice modulation and Navarasas.

#### **Course Outcomes:**

On successful completion of this course, students should be able to:

- Freely express improvisation in non-verbal communication.
- Shall hone good acting skills and be able to emote better.
- Be able to put up a theatre act and play a key role.
- Be able to differentiate good acting and understand the importance of good lyrics, stage crafting, music, dance, costume and lighting.

#### **Course Content:**



## UNIT – 1

### ***Working on Body:***

Body and its analysis. Understanding physical abilities ( Anga, Pratyanga and Upanga). Challenges of the body. Using body as metaphor and language. The class's bodies as a collective, an ensemble, a collaborative team.

## UNIT – 2

### ***Sound and Movement:***

Awareness of creating sound patterns, voice modulations, rhythm in speech and dialogues. Understanding the rhythm and patterns of movements like walking, framing, shaping, primitive and animal movements.

## UNIT – 3

### ***Characterization and Improvisation:***

Observation of people around. Getting into the role and living it. Developing a character from establishment (pace and rhythm). Improvisation techniques of body and mind.

## UNIT – 4

### ***Group work and Production:***

Develop a theme, concept or a play and include all the theatre skills, stage craft, costuming and put up an act. Choosing theme and characters.

### **Reference Books:**

1. All about Theatre – Off stage – Chris Hogget.
2. Rangadalli Anataranga – K V Subbanna
3. The Indian Theatre – Hemendranath Das Gupta.
4. A Practical handbook for an Actor – Milisa Bruder, ee Milchel Cohn, Madeleine Oliek et al, Zigler Publisher.

## **G. INDIAN CLASSICAL DANCE FORMS (Bharathanatyam, Kuchipudi ,Mohiniyattam)**

**Prerequisites: Background of classical dance training or any other dance forms.**

**Note: Non-classical dancers can also join.**

### **Course Objectives:**

- To develop an understanding about the Indian classical dance forms and its universal application.
- To be able to understand the fine nuances of Classical dance.
- To understand the importance of health through Indian classical dance, strengthen the body capacity.
- To understand mythology and its characters in Indian classical dance form through lessons of Abhinaya.

### **Course Outcomes:**

- To be able to identify and appreciate the classical dance forms.
- To be able to execute basics of Adavus with finesse.
- To be able to express through abhinaya.
- To be able to perform to perform the fundamentals in the chosen dance form.

**Course Content:****Unit 1**

An introduction to Indian classical dance forms: Bharatanatyam, Kuchipudi, Mohiniyattam.

**Unit 2**

Learning of Fundamentals: Exercises and Adavus- I (Bharathanatyam , Kuchipudi, Mohiniyattam).

**Unit 3**

Adavus –II (Bharathanatyam , Kuchipudi, Mohiniyattam)

**Unit 4**

Learn a basic composition in the chosen dance form.

**Reference Books:**

1. Indian classical dance forms –U S Krishna Rao,U K Chandrabhaga Devi
2. Classical Dances –Sonal Mansingh, Avinash Parischa
3. Kuchipudi – Sunil Kothari
4. Bharatanatyam An in depth study- Saroja vydyanathan
5. Mohiniyattam – Bharathi Shivaji

**H. PERCUSSION INSTRUMENT (TABLA AND MRIDANGAM)**

**Pre-requisites:** Students with background in Percussion instruments and knowledge of Rhythm/ Keen interest in studying Mridagam / Tabala.

**Course Objectives:**

- To understand the Rhythmology.
- To understand the importance of Laya, Taala.
- To be able to understand the fine finger techniques of playing the instrument.

**Course Outcomes:**

On successful completion of this course, students should be able to:

- To be able to set instrument to Sruthi.
- To be able to play the fundamentals on instrument.
- To be able to learn and perform a particular taala.

**Course Content:****UNIT 1**

1. Introduction to Musical Instruments
2. Percussion Instruments
3. Mridangam and its History

**UNIT 2**

1. Introduction to Tala System
2. Definitions of 5 jaathis and their recitation
3. Adi Talam and its various forms
4. Definitions and recitation of different gathers

**UNIT 3**

1. Tisra Jaathi
2. Khanda Jaathi
3. Misra jaathi
4. Sankeerna Jaathi

**UNIT 4**

1. Learning of Jathi Formation
2. Basic jathis
3. Jathis for Dance forms
4. Some Basic Definitions of Korvai, Teermanam etc.,

**Reference Books:**

1. Mridangam- An Indian Classical Percussion Drum – Shreejyanthi Gopal
2. Theory and practice of Tabala – Sadanand Naimpally.
3. Theory and practice of Mridangam – Dharmala Rama Murthy
4. The Art of the Indian Tabala – Srdjan Beronja.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH2070	SKILL DEVELOPMENT	RULO	0	0	2	2	3

**Note:** The students will have to undergo Skill Development course by Training and Placement cell of the University.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH2080	GENERAL PHYSICS LAB - II (PRACTICAL)	HC	0	0	3	3	5

**Prerequisites:** basic theory of general physics lab

**Course Objectives:**

- To achieve perfectness in experimental skills and the study of practical applications will bring more confidence and ability to develop and fabricate engineering and technical equipments.
- Design of circuits using new technology and latest components and to develop practical applications of engineering materials and use of principle in the right
- way to implement the modern technology.

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS02	PS03	PSO 4
M18PH2053	CO1	3	2	2	2	2	1	1	1			1		3		
	CO2	3	2	2	2	2	1	1	1			1		3		
	CO3	3	2	2	2	2	1	1	1			1		3		
	CO4	3	2	3	3	3	1	1	1			1		3		

**Course Outcomes:**

On Successful completion of this course, students shall be able to:

1. Infer the result of related experiments.
2. Employ the different tools and techniques to get the data/readings related to the experiments.
3. Explore the fundamental physics behind many scientific discoveries through hands on experience

**List of Experiments:**

1. Determination of difference in wavelengths of  $D_1$  and  $D_2$  lines using Michelson interferometer.
2. Active low pass and high pass filter using op-amp.
3. Determination of Fermi Energy of given conductor/semiconductor.
4. Experiment with GM counter.
5. Determination of Ferroelectric phase transition and verification Curie Weis law.
6. Measurement of thickness of thin wire using Laser source.
7. Determination of size of the particles using Laser.
8. Determination of velocity of ultrasonic waves in liquid.
9. Determination of Red berg constant.
10. Verification of photoelectric equation and determination photonic work function and Planck's constant.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./Wk.
M18PH2090	ATOMIC AND MOLECULAR PHYSICS (PRACTICAL)	H C	0	0	3	3	5

**Prerequisites:** basic theory of atomic physics lab

**Course Objectives:**

- To achieve perfectness in experimental skills and the study of practical applications will bring more confidence and ability to develop and fabricate engineering and technical equipments.
- Design of circuits using new technology and latest components and to develop practical applications of engineering materials and use of principle in the right way to implement the modern technology.

**Course Outcomes:**

On successful completion of this course, the student shall be able to:

Course Code	POS/COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS02	PS03	PS04
M18PH2053	CO1	3	2	2	2	2	1	1	1			1		3		
	CO2	3	2	2	2	2	1	1	1			1		3		
	CO3	3	2	2	2	2	1	1	1			1		3		
	CO4	3	2	3	3	3	1	1	1			1		3		

1. Verify the spectroscopic phenomenon of physics through experimentation.

2. Infer results of the experiments connected with interaction of electric and magnetic fields with atoms and molecules.
3. Distinguish the band, line and Raman spectrum through the experimentation.

### List of Experiments

1. CCD spectrometer to record absorption bands of Iodine molecule.
2. CCD spectrometer to record band spectrum of AlO.
3. Analysis of band spectrum of PN molecule.
4. Analysis of Rotational Raman spectrum of a molecule.
5. Determination of separation between etalon plates using Fabry-Perot interferometer experiments.
6. Zeeman effect.
7. Determination of  $e/m$  ratio of an electron by Millikan's oil drop experiment.
8. Study of intensity distribution of elliptically polarized light.
9. Study of elliptically polarized light using Babinet compensator.
10. Determination of thickness of mica sheet using Edser Butler Fringes.

## THIRD SEMESTER

Course Code	Course Title	Course Type	L	T	P	C	Hrs. / Wk .
M18PH3010	ELECTRODYNAMICS	HC	2	1	0	3	4

**Prerequisites:** Basics of electromagnetic theory

### Course Objective:

To introduce students the principles and applications of Electrostatics, Magneto-statics, Electrodynamics and Electromagnetic waves.

### Course Outcomes:

On the successful completion of this course, the student shall be able to:

1. Apply reasoning skills to model and solve problems related to electrostatics
2. Formulate problems within magnetostatics and stationary current distributions and solve.
3. Demonstrate the understanding of Faraday's laws and Maxwell's equations and physics concepts in Electrodynamics.
4. Derive expressions for the energy both for the electrostatic and magneto-static fields by using Poynting's theorem and Maxwell's equations.
5. Analysis and explain wave guides and Electromagnetic radiation by using concepts of electrostatics, magneto statics and Maxwell's equations.
6. Communicate scientific information in electrostatics, magneto statics, electrodynamics and electromagnetic radiation in oral, written, and graphical formats.

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PS01	PS02	PS03
M18PH1050	CO1	1	3	3		1	1	1			2	3	2	3
	CO2	2	3	3		1	1	1			2	3	2	3
	CO3	3	3	3		1	1	1			2	3	2	3
	CO4	4	3	3		1	1	1			2	3	2	3

## Course content

### Unit -1:

**Electrostatics:** Divergence and curl of electrostatic field, Gauss law in integral and differential forms, Poisson and Laplace equations, Boundary conditions and uniqueness theorem, electrostatic potential energy and energy density of a continuous charge distribution. Multipole expansion of the potential and energy of a localized charge distribution, monopole and dipole terms, electric field of a dipole, dipole-dipole interaction. Electrostatic fields in matter, polarization, macroscopic field equations, electrostatic energy in dielectric media. **12 hrs**

### Unit - 2:

**Magnetostatics:** Current density, continuity equation, magnetic field of a steady current, the divergence and curl of **B**, Ampere's law, magnetic vector potential, multipole expansion of vector potential of a localized current distribution, magnetic moment. Torques and forces on magnetic dipoles, effect of a magnetic field on atomic orbits. Magnetic fields in matter, macroscopic equations, magneto static boundary conditions, magnetic scalar potential. Energy in the magnetic field. **12 hrs**

### Unit -3:

**Electrodynamics:** Faraday law of induction, displacement current, Maxwell's equations. Vector and scalar potentials. Gauge transformations, Lorentz gauge, Coulomb gauge. Poynting's theorem and conservation of energy and momentum for a system of charged particles and electromagnetic fields.

**Electromagnetic Waves:** Plane waves in non-conducting and conducting medium, skin depth. Linear and circular polarizations. Reflection and refraction of plane waves at a plane interface, total internal reflection, reflection from a surface of a metal. **12 hrs**

### Unit 4:

**Wave guides:** Fields at the surface and within a conductor, cylindrical cavities and wave guides, modes in rectangular wave guide.

**Electromagnetic radiation:** Retarded Potentials. Radiation from an oscillating dipole, liner antenna. Lenard-Wiechert potentials, potentials for a charge in uniform motion, power radiated by an accelerated charge at low velocities, Larmor's formula, radiation from a charged particle with collinear velocity and acceleration, Bremsstrahlung radiation, radiation from a charged particle moving in a circular orbit, cyclotron and synchrotron radiation.

**Plasma Physics:** Plasma behavior in magnetic field, plasma as a conducting fluid-magneto hydrodynamics, magnetic confinement-Pinch effect. **12 hrs**

**References:**

1. Classical Electrodynamics: J.D.Jackson , Wiley Eastern Ltd., Bangalore (1978)
2. Introduction to Electrodynamics: D.J.Griffiths, Prentice Hall of India, Ltd., New Delhi (1995).
2. Electro magnetics: B.B. Laud. Wiley Eastern Ltd., Bangalore (1987)
3. Classical Electromagnetic Radiation: J.B. Marion, Academic press, NewYork (1968).
4. Classical Electrodynamics; S P Puri, Tata McGraw –Hill Publishing Company Ltd., New Delhi,(1990).

Course Code	Course Title	Course Type	L	T	P	C	Hrs./Wk.
M18PH3020	NANOSCIENCE AND TECHONOLOGY	HC	2	1	0	3	4

**Prerequisites:** Basics of electromagnetic theory

**Course Objectives:**

- To understand the fundamental concepts behind nanoscience and nanotechnology.
- To familiarize with various processing techniques available for synthesis of nanostructure materials.
- To acquire the knowledge of various nanomaterial characterization methods.
- To get familiarized with the various analytical techniques.
- To understand the properties of nanomaterials.

**Course Outcomes:**

On successful completion of this course, the student shall be able to:

1. Analyze the fundamental principles of nanotechnology and apply to different applications
2. Apply physics concepts to the nano-scale and non-continuum domain.
3. Demonstrate state-of-the-art nano-fabrication methods to prepare nano particles
4. Evaluate processing conditions to functional nanomaterials
5. Evaluate current constraints, such as regulatory, ethical, political, social and economical, encountered when solving problems in living systems.

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS011	PS02	PS03	PS04
M18PH3020	CO1	1	2	2	2	2	1	1	1				1		3	
	CO2	1	2	2	2	2	1	1	1				1		3	
	CO3	1	2	2	2	2	1	1	1				1		3	
	CO4	1	2	3	3	3	1	1	1				1		3	

**Course Content:****Unit – 1:****12 hrs**

NANOSCALE SYSTEMS: Length, energy, and time scales - Quantum confinement of electrons in semiconductor nanostructures: Quantum confinement in 3D, 2D, 1D and zero dimensional structures - Size effect and properties of nanostructures- Landauer Buttiker formalism for conduction in confined geometries - Top down and Bottom-up approach.

**Unit - 2:****12 hrs**

QUANTUM DOTS: Excitons and excitonic Bohr radius – difference between nanoparticles and quantum dots - Preparation through colloidal methods - Epitaxial methods- MOCVD and MBE growth of quantum dots - current-voltage characteristics - magneto tunneling measurements - spectroscopy of Quantum Dots: Absorption and emission spectra - photo luminescence spectrum - optical spectroscopy – linear and nonlinear optical spectroscopy.

**Unit – 3****12 hrs****SYNTHESIS OF NANOSTRUCTURE MATERIALS:**

Gas phase condensation – Vacuum deposition -Physical vapor deposition (PVD) - chemical vapor deposition ( CVD) – laser ablation- Sol-Gel- Ball milling –Electro deposition- electroless deposition – spray pyrolysis – plasma-based synthesis process (PSP) - hydrothermal synthesis.

**Unit – 4:****12 hrs**

**NANOTECHNOLOGY APPLICATIONS:** Applications of nanoparticles, quantum dots, nanotubes and nanowires for nanodevice fabrication – Single electron transistors, coulomb blockade effects in ultrasmall metallic tunnel junctions - nanoparticles based solar cells and quantum dots based white LEDs – CNT based transistors – principle of dip pen lithography.

**Reference Books:**

1. Nanotechnology|| G. Timp. Editor, AIP press, Springer-Verlag, New York, 1999
2. Nanostructured materials and nanotechnology “, Concise Edition, Editor:- Hari Singh Nalwa; Academic Press, USA (2002).
3. Hand book of Nanostructured Materials and Technology“, Vol.1-5, Editor:- Hari Singh Nalwa; Academic Press, USA (2000).
4. Hand book of Nanoscience, Engineering and Technology (The Electrical Engineering handbook series), Kluwer Publishers, 2002.
- 5 Sol-Gel Science||, C.J. Brinker and G.W. Scherrer, Academic Press, Boston (1994).
6. Nanoscale characterization of surfaces & interfaces, N John Dinardo, Weinheim Cambridge: Wiley-VCH, 2nd ed., 2000.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M19PH3030	NUCLEAR AND PARTICLE PHYSICS	HC	2	1	0	3	4

**Course Objectives:**

- To study the general properties of nucleus.
- To study the nuclear forces and nuclear reactions.
- To introduce the concept of elementary particles.



### Course Outcomes:

On successful completion of this course, the student shall be able to:

- Explain various types of nuclear reactions
- Construct nuclear and semi-conductor detectors
- Apply various models to study nuclear decay
- Apply basic laws of particle physics and macroscopic physics phenomena in determination of particle properties and properties of processes in the subatomic world

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS02	PS03	PS04
M19PH3030	CO1	3	3	2	3	2	1					2			3	
	CO2	3	3	2	3	2		1	1			2			3	
	CO3	3	3	1	3	3						2			3	
	CO4	3	3	1	2	2		1							3	

### Course Contents:

#### Unit 1

**Interaction of charged particles:** energy loss of heavy charged particles in matter, Bethe-Bloch formula, energy loss of fast electrons, Bremstrahlung.

**Interaction of gamma rays:** photo electric, Compton, and pair production processes. Gamma ray attenuation- attenuation coefficients, absorber, mass thickness, cross sections.

**Nuclear reactions:** cross section for a nuclear reaction, Differential cross section, Q-value of reaction, threshold energy, Direct and compound nuclear mechanisms, Bhor's independence hypothesis and experimental verification.

**Nuclear fission:** energy released in fission, neutron cycle in a thermal reactor and four factor formula.  
**12 hrs**

#### Unit 2

**Nuclear forces:** characteristics of nuclear forces, short range, saturation, charge independence and exchange characteristics, Ground state of deuteron, Relation between the range and the depth of the potential using square well potential, Yukawa's theory of nuclear forces (qualitative only)

**Nuclear detectors:** scintillation detectors- NaI(Tl), plastic scintillation- scintillation spectrometer.

**Semiconductor detectors:** Surface barrier detectors, Li ion drift detectors, relation between applied voltage and the depletion region in junction detectors, counter telescopes, particle identification, and position sensitive detector.

**12 hrs**

#### Unit 3

##### Nuclear models and nuclear decay:

**Liquid drop model:** Semi empirical mass formula, stability of nuclei against beta decay, mass parabola.

**Fermi gas model:** Kinetic energy for the ground state, asymmetry energy.

**Shell model:** evidence for magic numbers, prediction of energy levels in an infinite square well potential, spin orbit interaction, prediction of ground state spin parity and magnetic moment of odd nuclei, Schmidt limit.

**Beta decay:** Fermi's theory of beta decay, curie plots and ft values, selection rules.

**Gamma decay:** Multipolarity of gamma rays, selection rules, internal conversion (qualitative only).

**12 hrs**

#### Unit 4

**Elementary particle physics:** types of interactions between elementary particles, hadrons and leptons, detection of neutrinos.

**Symmetries and conservation laws:** conservation of energy, momentum, angular momentum, charge and isospin, parity symmetry, violation of parity in weak interactions, lepton number conservation, lepton family and three generations of neutrinos. Conjugation symmetry, CP violation in weak interactions.

Strange particles, conservation of strangeness in strong interactions, Baryon number conservation, Gell-Mann Nishijima formula, eight fold way (qualitative only), Quark model, quark content of baryons and mesons, color degree of freedom, standard model (qualitative only). **12 hrs**

#### References:

1. Introduction to Nuclear Physics H. Enge: Addison Wesley, 1971.
2. Atomic and Nuclear Physics, S. N. Goshal vol II 2000.
3. Introductory Nuclear Physics Kenneth S. Krane: John Wiley and Sons, 1987.
4. The Atomic Nucleus Evans R.D. : Tata Mc. Graw hill, 1955.
5. Nuclear Physics, R R Roy and Nigam: Wiley-eastern Ltd 1983.
6. Nuclear physics an introduction, S.B. Patel: New age international (P) limited 2000.
7. Radiation Detection and Measurements, G.F. Knoll: 3<sup>rd</sup> edition, John Wiley and sons, 2000.
8. Nuclear Radiation Detectors, S.S. Kapoor and V.S Ramamurthy: Wiley and sons. Introduction to High Energy Physics D.H. Perkins: Addison Wesley, London, 2000.
9. Introduction to Elementary Particles, D.Griffiths: John Wiley 1984.
10. Nuclear Interactions, S.de Benedetti: John Wiley, New York, 1964.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH3030	ELECTRONICS - II (ELECTRONIC COMMUNICATION SYSTEMS)	HC	2	1	0	3	4

**Prerequisites:** Basics of electronics 1

#### Course Objectives:

1. To understand the fundamental concepts of communication systems.
2. To understand and compare different analog modulation schemes.
3. To understand and compare different digital modulation schemes.
4. To understand the design tradeoffs and performance of communications systems.

#### Course Outcomes:

On successful completion of this course, the student shall be able to:

1. Analyze the working of amplitude and frequency modulated communication systems.
2. Analyze various modulation schemes in digital communication system
3. Distinguish between Analog and Digital Communication system and analyze various sampling methods and its reconstruction.
4. Construct various channel coding and decoding schemes such as Hadamard code, Hamming code, Cyclic Codes, Convolution coding

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS0 1	PS0 2	PS0 3	PS0 4
M18PH3030	CO1	2	3							2		1				3
	CO2	2	3	2	1	2						2				3
	CO3	2	3	2								2				3
	CO4	2	3	2	1	2						2				3

### Course Content:

#### Unit - 1: Analog communication systems

AM Transmitters and Receivers: Generation of AM, low level and high level modulation, comparison of levels, AM transmitter block diagram, DSB S/C modulator. AM Receiver: Tuned radio frequency (TRF) receiver. Super heterodyne receiver, RF section and characteristics, mixers, frequency changing and tracking, IF rejection and IF amplifiers. Detection and automatic gain control (AGC), AM receiver characteristics.

**12 hrs**

#### Unit - 2:

FM Transmitters and Receivers FM Transmitters: Basic requirements and generation of FM, FM Modulation methods: Direct methods, Variable capacitor Modulator, Varactor Diode Modulator, FET Reactance Modulator, Transistor Reactance Modulator, Pre-emphasis, Direct FM modulator, FM Receivers: Limiters, single and double-tuned demodulators, balanced slope detector, Foster-Seely or Phase Discriminator, De-emphasis, ratio Detector, Block diagram of FM Receivers, RF Amplifiers, FM Receiver characteristics.

**12 hrs**

#### Unit - 3: Digital communication systems

Analog to Digital Conversion Noisy communications channels, The sampling Theorem, low pass signals and band pass signals, pulse Amplitude modulation, channel bandwidth for a PAM signal, Natural sampling, Flat top sampling, signal recovery & holding, Quantization of signal, Quantization error, pulse code modulation (PCM), Delta Modulation, adaptive delta modulation. Digital Modulation Techniques: Binary phase shift keying, differential phase shift keying, differential encoded PSK, QPSK, Quadrate Amplitude shift keying (QASK) Binary frequency shift keying.

**12 hrs**

#### Unit - 4:

Information Coding and Decoding: Coding for error detection and correction, Block coding – coding, anticoding, Hadamard code, Hamming code, Cyclic Codes, Convolution coding and decoding, Viterbi algorithm, Shannon Fano and Hoffman Codes.

**12 hrs**

### Reference Books:

1. Principles of Communication Systems – H Taub& D. Schilling, GautamSahe. TMH, 2007 3rd Edition.
2. Principles of Communication Systems - Simon Haykin. John Wiley, 2r" Edition,
3. Electronics & Communication System - George Kennedy and Bernard Davis, 4th Edition TMH 2009
4. Analog Communications- KN Hari Bhat & Ganesh Rao, Pearson Publications, 2nd Edition 2008.
5. Analog and Digital Communication – K. Sam Shanmugam, Willey ,2005
6. Electronics Communication Systems-Fundamentals through Advanced-Wayne Tomasi, 5th Edition, 2009, PHI.
7. Lathi "Modern Digital and Analog Communication Systems," Oxford University Press.

8. B. Sklar, "Digital Communications: Fundamentals and Applications," Pearson Education.
9. S. Haykin, "Digital Communication," John Wiley.
10. R.P. Singh and S.D. Sapre, "Communication Systems: Analog and Digital," Tata McGraw-Hill.
11. Digital Communications - John G. Proakis .Masoudsalehi – 5th Edition, McGraw-Hill, 2008.
12. Digital Communication - Simon Haykin, Jon Wiley, 2005.
13. Digital Communications - Ian A. Glover, Peter M. Grant, Edition, Pearson Edu., 2008.
14. Communication Systems-B.P. Lathi, BS Publication, 2006.
15. Principles of communication systems - Herbert Taub. Donald L Schiling, Goutam Sana, 3rd Edition, Mc. Graw-Hill, 2008.
16. Digital and Analog Communicator Systems - Sam Shanmugam, John Wiley, 2005.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH3042	CONDENSED MATTER PHYSICS –II	SC	3	1	0	4	5

**Prerequisites:** Knowledge on condensed matter physics 1

#### Course Objectives:

The course is to understand the basic knowledge on magnetic, dielectric and electric properties of material, types of magnetic and dielectric materials and their applications.

#### Course Outcomes:

On successful completion of this course, the student shall be able to:

1. Differentiate types of magnetic materials.
2. Analyze ESR and EPR spectral data.
3. Understand classical and quantum theory of dielectrics.
4. Classify ferroelectric crystals.

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS0 2	PS0 3	PSO 4
M18PH3042	CO1	3	2	1	2	2			1	1		2		3		
	CO2	3	3	3	3	3	1	1	1	1		2		3		
	CO3	3	2	2	2	2			1	1				3		
	CO4	2	2	1	1	1			1	1		2		3		

#### Course Content:

##### Unit - 1:

**Ferromagnetism:** Review of Weiss theory of ferromagnetism, its successes and failures, Heisenberg exchange interaction, exchange integral, exchange energy, Ising model, Spin waves (one dimensional case only), quantization of spin waves and magnons, density of modes, thermal excitation of magnons and Bloch  $T^{3/2}$  law, specific heat using spin wave theory. Band theory of ferromagnetism. Ferromagnetic domains, hysteresis curve, magneto crystalline anisotropy energy, Bloch wall.

**Ant ferromagnetism:** Characteristic property of anti-ferromagnetic substance, Neutron diffraction experiment. Two sub-lattice model molecular field theory of anti-ferromagnetism, Neel temperature, Susceptibility below and above Neel temperature.

**Ferrimagnetism:** Ferrimagnetic order, ferrites, Curie temperature and susceptibility of ferrimagnets.

**12 hrs**

**Unit - 2:**

**Magnetic Resonance:** Basic principles of paramagnetic resonance, spin-spin and spin-lattice relaxation, susceptibility in a.c. magnetic field power absorption, equations of Bloch, steady state solutions, determination of g-factor, line width and spin-lattice relaxation time, paramagnetic resonance and nuclear magnetic resonance. Effect of crystal field on energy levels of magnetic ions (qualitative). Spin- Hamiltonian, zero field splitting. **12 hrs**

**Unit - 3:**

**Dielectrics:** Review of basic formulae, dielectric constant and polarizability, local field, Clausius-Mossotti relation, polarization catastrophe. Sources of polarizability, Dipolar polarizability: dipolar dispersion, Debye's equations, dielectric loss, dipolar polarization in solids, dielectric relaxation. Ionic polarizability. Electronic polarizability: classical treatment, quantum theory, interband transitions in solids. **12 hrs**

**Unit - 4:**

**Ferroelectrics:** General properties of ferroelectrics, classification and properties of representative ferroelectric crystals, dipole theory of ferroelectricity, dielectric constant near Curie temperature, microscopic source of ferroelectricity, Lydane-Sachs-Teller relation and its implications, thermodynamics of ferroelectric phase transition, ferroelectric domains, Piezoelectricity and its applications. **12 hrs**

**Reference Books:**

1. The Physical Principles of Magnetism : A. H. Morrish, John Wiley & sons, New York (1965)
2. Solid State Physics : A. J. Dekker, Macmillan India Ltd., Bangalore (1981)
3. Introduction to Solid State Physics : 5th Edn C. Kittel, Wiley Eastern Ltd., Bangalore (1976)
4. Elementary Solid State Physics : M. A. Omar, Addison-Wesley Pvt. Ltd., New Delhi (2000)
5. Introduction to Magnetic Resonance: A. Carrington and A. D. Mclachlan, Harper & Row, New York, (1967).
6. Elements of Solid State Physics (2nd Ed): J.P. Srivastava, PHI Learning Pvt. Ltd., New Delhi (2009)

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH3043	MATERIAL SCIENCE - II	SC	3	1	0	4	5

**Prerequisites:** Knowledge on material science-1

**Course Objectives:**

This course provides the knowledge about the processing, characterization and testing of Polymers, ceramics and glass materials.

**Course Outcomes:**

On successful completion of this course, the student shall be able to:

1. Synthesize the polymers, ceramics and glasses.
2. Explain the applications of composite material related to polymers, ceramics and glasses
3. Analyze the Properties of polymers, ceramics and glasses
4. Analyze the defects and inabilities of polymers, ceramics and glasses for industrial applications.

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO1 0	PO1 1	PS01	PS02	PS0 3	PSO 4
M18PH3043	CO1	3	3		1	1	1									3
	CO2	3	2	2	2	2			1							3
	CO3	3	2	1	2	2	1		1							3
	CO4	3	2	2	2	2			1							3

### Course Content:

#### Unit-1:

**Elements of Polymer Science:** Monomers- Polymers- Classification of polymers Synthesis of polymers- chain polymerization, step Polymerization, industrial polymerization methods. Average molecular weight-weight, number and viscosity, size of polymer molecule. Microstructure of polymers- chemical, geometric, random alternating, block and graft polymers, stereo regular polymers. Phase transition- polymer melting and glass transition; polymer crystallinity- degree of crystallinity, crystallization and stereo isomerism. Processing of Plastic Materials- inoculating- compression, injection blow, extrusion, spinning.

**12 hrs**

#### Unit-2:

**Ceramics and Glasses:** Ceramics and their structure- Silicate structure, Preparation-Forming and thermal treatments, Classification of ceramics- traditional and engineering. Dielectric, ferroelectric and piezoelectric properties of ceramics with specific examples, Ceramic magnets, Mechanical properties- strength, toughness. Fatigue failure, abrasion. Basic refractory materials.

**Glasses:** Preparation and structure, Types of glasses- borates silicate, oxide, metallic and semiconducting glasses; tempered glass and chemically strengthened glass.

**12 hrs**

#### Unit-3:

**Composite Materials:** General Introduction, matrix Materials- polymer, metals, ceramics, Reinforcing materials- fibers, particles. Concrete-concrete making materials, structure, composition. properties and applications. Polymer-concrete composites, fabrication, structure, interface, properties, applications of polymer matrix composites, metal matrix composites, ceramic matrix composites and carbon fiber composites, wood-plastic composites, dispersion strengthened. Particle reinforced, fiber and laminate reinforced composites with fabrication, interface, properties and applications.

**12 hrs**

**Unit-4: Testing of Materials:** Mechanical Testing - Universal Testing Machine. Hardness- Brinell, Vicker and Rockwell, impact testing and Fatigue Testing. Optical Microscopy- Metallurgical Microscopes- sample preparation and grain size Measurements. Electron microscopy-Transmission microscopy (TEM), scanning microscopy (SEM)- principle, sample preparation techniques and applications. non Destructive Testing- Ultrasonic Testing, X-ray radiography. Neutron radiography.

**12 hrs**

### Reference Books:

1. Textbook of Polymer Science. **Fred. W. Billmeyer** John Wiley & Sons, Inc. (1984)
2. Polymer Science, **V.R. Gowariker, N. V. Vistrwanathan, Jayadev Shreedhar**, Wiley Eastern (1937)
3. Electronic properties of Materials- **Rolf E. Hummel, Springer Verlag**, Springer Verlag (1985)
4. Foundations of Materials Science and Engineering- **William F. Smith**, McGraw Hill international Editions, (1988)
5. Elements of Materials Science and Engineering. Lawrence **H. Van Vlack**, Addison Wesley (1975)
6. Introduction to Ceramics- **W D Kingery, H K Bower and U R Uhlman**, John Wiley (1960)
7. Ultrasonic **B. Carlin**, Mc. Graw Hill (1950).
8. Principles of Neutron Radiography- **N D Tyufyakav and A S Shtan**, Amerind Publishers (1979)
9. Applied X-rays- **George L Clark**, Mc. Graw Hill, (1955)
10. Testing of Metallic Materials— **AVK Suryanarayan**, Prentice Hall India, (1990)
11. Physical Metallurgy Part I, **R W Cahn and P Haasen** (Ed), North.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH3051	ELECTRONICS - III (LINEAR INTEGRATED CIRCUITS)	SC	3	1	0	4	5

### Course Outcomes:

On successful completion of this course, the student shall be able to:

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS02	PS03	PS04
M18PH3053	CO1	3	3		1	1	1									3
	CO2	3	2	2	2	2			1							3
	CO3	3	2	1	2	2	1		1							3
	CO4	3	2	2	2	2			1							3

1. Characterize optical and dielectric properties of metals & non-metals through the spectroscopic techniques.
2. Analyze the properties like Luminescence and photoconductivity of metals & non-metals.
3. Design the metals and non-metals having the properties of stress & strain, elasticity, Plastic deformation.
4. Explain the effect of temperature on the properties of metals and alloys.
5. Explain to use the techniques of measurements & instruments.

### Course Content:

#### Unit-1:

**Optical and dielectric properties of materials:** Theory of electronic polarization and optical absorption, ionic polarization, orientationl polarization. Optical phonon model in an ionic crystal;

Interaction of electromagnetic waves with optical modes, polarization, Dispersion curves of transverse optical (TO) phonon and optical photon in a diatomic ionic crystal, LST relation; Metal-insulator transition. UV-VIS, IR, FTIR and Raman spectroscopy. Optical properties of metals & nonmetals- Luminescence, photoconductivity. **12hrs**

#### Unit -2:

**Electrical properties of crystalline, nanocrystalline and polymeric materials:** Resistivity variation in metals, alloys, semiconductors and nanocrystalline materials, electrical conduction in ionic ceramics, clay materials and conducting polymers. Two-probe and four probe techniques, DC and AC conductivity measurements.

**Mechanical Properties of metals and ceramics:** Concepts of stress & strain, stress-strain behavior, elasticity, Plastic deformation, Hardness-Knoop & Vicker's hardness test.

**12 hrs**

#### Unit - 3:

**Thermal properties of metals & alloys:** Temperature effects on the intensities of Bragg reflections. Influence of temperature on diffraction of X-rays: Normal coordinates of lattice vibration and X-ray scattering from a vibrating lattice and origin of thermal diffuse spots. First order TDS. Debye-Waller factor' Debye's method of calculating isotropic temperature factor for a cubic crystal. DTA, TGA, DSC (Outline only). Annealing processes, Heat treatment of steels, mechanism of hardening. Quenching, thermal stresses.

**12 hrs**

#### Unit - 4:

**Structure - Property correlation,** Correlation of structure with physical properties of materials, application prospects of materials in different areas.

**Basic concepts of measurements & instruments:** Static characteristics of instruments, accuracy & precision, sensitivity, reproducibility, errors, Transducers, classification & selection criteria, principles of piezoelectric, photoelectric, thermoelectric transducers, resistance temperature transducers (RTD), Thermistor, strain gauge, load cells, LVDT Electronic instruments for measurement, Digital voltmeter, principles of electronic multimeter, digital multimeter, Q-meter, Electronic LCR meter, Frequency & time interval counters.

**12 hrs**

#### Reference Books:

1. Introduction to Ceramics by W. D. Kingery, H. K. Bowen and D. R. Uhlmann, John Wiley & Sons.
2. Diffraction analysis of the microstructure of materials by E. J. Mittemeijere and P. Scardi, Springer
3. Materials Science & Engineering by William D. Callister, John Wiley & Sons, Inc.
4. Modern techniques of surface science by D. P. Woodruff & T. A. Delchar, Cambridge University Press.
5. X-ray spectroscopy by B. K. Agarwal, Springer-Verlag.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH3060	ASTROPHYSICS	OE	4	0	0	4	4

**Prerequisites:** fundamental Knowledge on Astrophysics



**Course Objectives:**

- To study the astrophysical universe, ranging from solar system objects through stars, to galaxies and the structure of the universe as a whole.
- To understand the principles and methods of modern astrophysics.

**Course Outcomes:**

On successful completion of this course, the student shall be able to:

1. Describe the laws that govern the astrophysical phenomena.
2. Explain the nature and properties of compact astrophysical objects.
3. Elaborate astrophysical observations and measurements.
4. Discuss the principles of formation of the Planets and Comets.

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS0 2	PS0 3	PS0 4
M18PH3060	CO1	3	3	2	1	3	1	1			2	1	3			
	CO2	3	3	1	3	3						3	3			
	CO3	3	3	1			1				2		3			
	CO4	3	3	1				1				1	3			

**Course Content:****Course Content:****Unit -1:**

**Basic concepts of Astronomy:** Co-ordinate system, Time system-Solar and Sidereal times, Apparent and Absolute magnitudes, Trigonometric Parallax, Atmospheric extinction, Optical telescopes and their characteristics, Modern Optical telescopes, Astronomical Instruments – Photometer, Photographic plates, Spectrographs, Charge Coupled Detector.

12 hrs

**Unit – 2:**

**Stellar properties:** Observational properties of stars – spectral and luminosity classification of stars- H-R Diagram, Saha Equation, , Star Formation - Jeans mass, Jeans Length and Free fall timescale, Main

Sequence Evolution, Mass- luminosity relation, White Dwarfs – Chandrasekhar's Limit, Neutron Stars, Pulsars, Supernovae, Stellar Black holes.

12 hrs

**Unit - 3:**

**Solar system:** Overview of Sun, Solar Interior structure- Core, Radiative zone and Convective Zone, solar atmosphere-photosphere, Chromospheres, Properties of Interior planets and exterior planets satellites of planets, Kuiper Belt objects, Oort Cloud, Theories of formation of the solar system.12 hrs

**Unit – 4:**

**Stellar structure:** Hydrostatic Equilibrium, Mass conservation, Luminosity gradient equation, Temperature gradient Equations, Lane – Emden equation for polytropic stars and its physical solution, estimates of central pressure and temperature, Radiation pressure, equation for generation and luminosity, equation of temperature gradient for radiative and convective equilibrium, Schwarzschild criterion, gas pressure and radiation pressure, Linear Model and its properties, Volt – Russell theorem,

**Reference Books:**

1. Ostlie and Carroll: Introduction to Modern Astrophysics, Addison Wesley (II Edition), 1997
2. Kristian Rohlif : Tools of Radio Astronomy, Springer
3. John D. Krauss : Radio Astronomy, II Edition, Signet.
4. F. Shu : The Physical Universe, University So Press, 1987.
5. M. Schwarzschild : Structure and Evolution of Stars, Dover.
6. R. Kippenhahn and Weigert A.: Stellar Structure and Evolution, Springer Verlag, 1990.
7. C.J. Hansen and Kawaler S.D.: Stellar Interiors: Physical Principles, Structure and Evolution, Springer Verlag, 1994.
8. M. S. Longair: High Energy Astrophysics, CUP.
9. Kitchin C R : Stars, Nebulae and the Interstellar Medium, Taylor and Francis Group, 1987.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./Wk.
M18PH3070	GENERAL PHYSICS LAB –III (PRACTICAL)	HC	0	0	3	3	5

**Prerequisites:** Knowledge on General Physics lab-II

**Course Objectives:**

To introduce the basic concepts of physics through hands on experience and impart experimental skill to students

**Course Outcomes:**

On successful completion of this course, the student shall be able to:

1. Conduct the experiments related to different Physics laws and theories.
2. Employ the different tools and techniques to get the data/readings related to the experiments.
3. Verify the fundamental physics behind many scientific discoveries through hands on experimentation.

**Course Content:**

Course Code	POS/COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS02	PS03	PS04
M18PH3070	CO1	3	3	2	1	3	1	1			2	1	3			
	CO2	3	3	1	3	3						3	3			
	CO3	3	3	1			1				2		3			
	CO4	3	3	1				1				1	3			

**LIST OF EXPERIMENTS:**

1. Hall effect experiment:- Determination of charge carrier density.
2. Spatial and temporal coherence of He-Ne laser.
3. Verification of Mallu's law.
4. Experiments with lasers and reflection grating.
5. To photograph the spectra of Fe (standard) and Cu arc using CDS spectrograph and to determine the wavelengths of Cu spectrum using Hartman formula.
6. Determination of laser light wavelength using Lloyd's mirror.
7. Determination of Numerical aperture in an optical cable.
8. Laser light attenuation in an optical cable.

9. Refractive index of liquids/solids using laser light.
10. Diffraction of laser light through two closely spaced circular apertures.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./Wk.
M18PH3081	ELECTRONICS LAB (PRACITICAL)	SC	0	0	3	3	5

**Prerequisites:** Knowledge on basic electronics

### Course Objectives:

The study of this course aims to:

1. Provide a strong foundation on Linear Circuits.
2. Familiarize students with applications of various IC's.
3. Have a broad coverage in the field that is relevant for engineers to design linear circuits using Op-amps.
4. Familiarize the conversion of data from Analog to Digital and Digital to Analog.

### Course Outcomes:

On successful completion of this course, the student shall be able to:

1. Analyze the functioning of basic electronic circuits of AM and FM modulation and demodulation through experimentation using discrete electronic components.
2. Verify sampling theorem by experiment.
3. Verify different modulation and demodulation techniques through experimentation.
4. Draw the outputs of various angle modulation and demodulation systems
5. Verify the outputs of ASK, FSK, PSK circuits.

### Course Content:

Course Code	POS/COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS02	PS03	PS04
M18PH3081	CO1	3	3	2	1	3	1	1			2	1	3			
	CO2	3	3	1	3	3						3	3			
	CO3	3	3	1			1				2		3			
	CO4	3	3	1				1				1	3			

### List of Experiments:

#### Analog and Digital communication lab

1. Amplitude modulation and demodulation.
2. DSB SC modulation and demodulation.
3. SSB SC modulation and demodulation.
4. Frequency modulation and demodulation.
5. Pre Emphasis-De Emphasis circuits.
6. Verification of sampling theorem.
7. PAM generation and reconstruction.
8. PWM AND PPM: generation and reconstruction
9. Delta and Adaptive delta modulation.

10. TDM of two band limited signals.
11. ASK generation and detection.
12. FSK generation and detection.
13. PSK generation and detection.
14. Line coding and decoding.

**LIC Lab:**

1. OP-AMP as square wave generator.
2. Schmitt trigger.
3. Voltage regulator
4. UJT relaxation oscillator
5. OP-AMP as active integrator and differentiator.
6. Design and test the operation of 4 Bit DAC using R-2R ladder network and OP-AMP MA741.
7. Design a second order Butterworth active low pass filter and high pass filter.
8. Design Schmitt trigger and test the circuit for the given values of UTP and LTP using OP-AMP MA741.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./Wk.
M19PH3082	CONDENSED MATTER PHYSICS LAB (PRACTCAL)	SC	0	0	3	3	5

**Course Objectives:**

1. To make the student familiarize with the basics of materials.
2. To enable the student to explore the concepts involved in the X-ray diffraction.
3. To make the student understand the basic concepts in absorption and Infrared spectroscopy.
4. To allow the student to understand the fundamentals of Hysteresis

**Course Outcomes:**

On successful completion of this course, the student shall be able to:

1. Identify the phase and phase purity of the crystal.
2. Determine particle size, stress and strain using PXRD.
3. Analyse of Thermoluminescence glow curve using ORIGIN software.
4. Determine of Curie temperature using B-H curve of a Ferromagnetic material (both hard and soft).
5. Calibrate of electromagnet and magnetic susceptibility determination of magnetic salts ( $\text{MnSO}_4$ ,  $\text{MnCl}_2$ ) by Quincke's method.
6. Study I-V characteristics of any given materials.

**Course Content:**

Course Code	POS/COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS02	PS03	PS04
M18PH3082	CO1	3	3	2	1	3	1	1			2	1	3			
	CO2	3	3	1	3	3						3	3			
	CO3	3	3	1			1				2		3			
	CO4	3	3	1				1				1	3			

**LIST OF EXPERIMENTS:**

1. Analyzing and determining the lattice parameters (h k l) values of FCC crystals by X-ray powder Diffractogram data.
2. Analyzing and determining the lattice parameters (h k l) values of BCC crystals by X-ray powder Diffractogram data.
3. Determination of particle size, stress and strain using PXRD data.
4. Synthesis of metal nanoparticles by solution combustion technique.
5. Determination of Reitveld refinement parameter using full proof suit software.
6. Determination of energy gap by using absorption spectra.
7. Analysis and estimation of kinetic parameters of Thermoluminescence glow curves.
8. Determination of Curie temperature using B-H curve of a Ferromagnetic material (both hard and soft).
9. Estimation of CIE coordinates of nanophosphor samples.
10. Study of I-V characteristics of semiconducting material by using Keithley source meter.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M19PH3083	MATERIAL SCIENCE LAB (PRACTICAL)	SC	0	0	3	3	5

**Course Objectives:**

1. To make the student familiarize with the basics of materials science.
2. To enable the student to explore the concepts involved in the X-ray diffraction.
3. To make the student understand the basic concepts in absorption and Infrared spectroscopy.
4. To allow the student to understand the fundamentals of Hall Effect and Hysteresis.

**Course Outcomes:**

On successful completion of this course, the student shall be able to:

1. Demonstrate different experimental techniques.
2. Experiment with X – ray diffractometer to determine the phase of the nano materials.
3. Analyze the thermoluminescence glow curves of different phosphor materials
4. Test I-V properties of thin films through experiment.

**Course Content:**

Course Code	POS/ COs	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PS01	PS02	PS03	PS04
M18PH3083	CO1	3	3	2	1	3	1	1			2	1	3			
	CO2	3	3	1	3	3						3	3			
	CO3	3	3	1			1				2		3			
	CO4	3	3	1				1				1	3			

**LIST OF EXPERIMENTS:**

1. Synthesis of nanomaterial by solution combustion technique.
2. Synthesis of metal nanoparticles by sonochemical method.
3. Determination of particle size and lattice strain of an unknown powder specimen applying Scherer Equation.
4. Determination of energy gap by using absorption spectra (UV-visible spectrometer).
5. Analysis and estimation of kinetic parameters of Thermo luminescence glow curves.
6. B-H curve of a Ferromagnetic material (both hard and soft).
7. Study of I-V characteristics of given material by Keithley instrument.
8. Analysis of single crystal rotation photograph.
9. Study of PL spectrum and determination of CIE coordinate values of nanophosphors samples.
10. Determination of lattice parameters values of FCC crystals by X-ray powder Diffractogram data.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH4010	MAJOR PROJECT	HC	0	0	10	10	20

**Course Objective:**

To carry out the research under the guidance of supervisor and in the process learn the techniques of research.

**Course Outcomes:**

On successful completion of the project, the student shall be able to:

1. Familiarize with literature search
2. Conduct the experiments related to research and formulate computational techniques
3. Interpret the scientific data.
4. Write report and defend the research findings.

**Project:**

Each student will choose the topic of research particularly from any area of soft cores studied and work under the guidance of allocated faculty member. The project shall preferably be application oriented or industry need based that could be useful to the society. In case of industry need base project the student may opt co-supervisor from the concerned industry. The student will have to make a preliminary survey of research done in broad area of his/her area of interest and decide on the topic in consultation with his/her supervisor(s). The project work floated should be completed within 16 weeks and project report has to be submitted within the stipulated date by the University/ within 18 weeks whichever is earlier. The student has to meet the concerned supervisor(s) frequently to seek guidance and also to produce the progress of the work being carried out. The student should also submit progress report during 5<sup>th</sup> week and 10<sup>th</sup> week of the beginning of the semester and final draft report with findings by 15<sup>th</sup> week. After the completion of the project the student shall submit project report in the form of dissertation on a specified date by the School.

Course Code	Course Title	Course Type	L	T	P	C	Hrs./ Wk.
M18PH4020	MOOC/SWAYAM/ INTERNSHIP	RULO	0	0	4	4	4

**Note:** Students shall choose to take up any online course of four credits as guided by the school or shall have to undergo internship of four weeks duration, the details of which are provided here under.

#### **MOOC/ SWAYAM:**

Globally, MOOC (Massive Open Online Course) platforms are gaining much popularity. Considering the popularity and relevance of MOOCs, Government of India has also launched an indigenous platform, SWAYAM. SWAYAM (Study Webs of Active Learning for Young Aspiring Minds) is basically an integrated MOOCs platform for distance education that is aimed at offering all the courses from school level (Class IX) to post-graduation level. The platform has been developed collaboratively by MHRD (Ministry of Human Resource Development) and AICTE (All India Council for Technical Education) with the help of Microsoft and is capable of hosting 2,000 courses.

A student shall register and successfully complete any of the courses available on SWAYAM.

Student shall inform the MOOC/SWAYAM coordinator of the school about the course to which he/she has enrolled. The minimum duration of the course shall be not less than 40 hours and of 4 credits. The student should submit the certificate issued by the SWAYAM to the MOOC/SWAYAM coordinator of the school, the grades obtained in the course shall be forwarded to concerned authority of the University.

**Internship:** Minimum of four weeks duration internship should be carried out by the student either in industry or in an R&D organization, including educational institutes with excellent research culture. In case, if a student is unable to secure internship either in industry or in an R&D organization, a project may be carried out within the university. The student is expected to submit a formal report at the end of the internship programme. The student shall be awarded the marks for internship based on the (a) presentation and (b) comprehensive viva by the panel of examiners constituted by the school.

## **Career Development and Placement**

Having a degree will open doors to the world of opportunities for you. But Employers are looking for much more than just a degree. They want graduates who stand out from the crowd and exhibit real life skills that can be applied to their organizations. Examples of such popular skills employers look for include:

1. Willingness to learn
2. Self-motivation
3. Teamwork
4. Communication skills and application of these skills to real scenarios
5. Requirement of gathering, design and analysis, development and testing skills
6. Analytical and Technical skills
7. Computer skills
8. Internet searching skills
9. Information consolidation and presentation skills
10. Role play
11. Group discussion, and so on

REVA University, therefore, has given utmost importance to develop these skills through variety of training programs and such other activities that induce the said skills among all students. A full-fledged Career Counseling and Placement division, namely Career Development Center (CDC) headed by well experienced senior Professor and Dean and supported by dynamic trainers, counselors and placement officers and other efficient supportive team does handle all aspects of Internships and placements for the students of REVA University. The prime objective of the CDC is to liaison between REVA graduating students and industries by providing a common platform where the prospective employer companies can identify suitable candidates for placement in their respective organization. The CDC organizes pre-placement training by professionals and also arranges expert talks to our students. It facilitates students to career guidance and improve their employability. In addition, CDC forms teams to perform mock interviews. It makes you to enjoy working with such teams and learn many things apart from working together in a team. It also makes you to participate in various student clubs which helps in developing team culture, variety of job skills and overall personality.

The need of the hour in the field of Physics is knowledge in the subject, but also the skill to do the job proficiently, team spirit and a flavor of innovation. This kept in focus, the CDC has designed the training process, which will commence from second semester along with the curriculum. Special coaching in personality development, career building, English proficiency, reasoning, puzzles, and communication skills to every student of REVA University is given with utmost care. The process involves continuous training and monitoring the students to develop their soft skills including interpersonal skills that will fetch them a job of repute in the area of his / her interest and march



forward to make better career. The School of Applied sciences also has emphasized subject based skill training through lab practice, internship, project work, industry interaction and many such skilling techniques. The students during their day-to-day studies are made to practice these skill techniques as these are inbuilt in the course curriculum. Concerned teachers also continuously guide and monitor the progress of students.

The University has also established University-Industry Interaction and Skill Development Centre headed by a Senior Professor & Director to facilitate skill related training to REVA students and other unemployed students around REVA campus. The center conducts variety of skill development programs to students to suite to their career opportunities. Through this skill development center the students shall compulsorily complete at least two skill / certification based programs before the completion of their degree. The University has collaborations with Industries, Corporate training organizations, research institutions and Government agencies like NSDC (National Skill Development Corporation) to conduct certification programs. REVA University has been recognized as a Centre of Skill Development and Training by NSDC (National Skill Development Corporation) under Pradhan Mantri Kaushal Vikas Yojana.

The University has also signed MOUs with Multi-National Companies, research institutions, and universities abroad to facilitate greater opportunities of employability, students' exchange programs for higher learning and for conducting certification programs.

## **DO'S AND DON'TS**

### **DO'S**

1. Maintain discipline and respect the rules and regulations of the university
2. Be regular and punctual to classes
3. Study regularly and submit assignments on time
4. Be respectful to your colleagues/friends and hostel staff/management.
5. Read the notice board (both at your college and the hostel) regularly.
6. Utilize your Personal Computer for educational purpose only.
7. Follow the code of conduct.
8. Visit Health Center on the campus whenever you are unwell.
9. Be security conscious and take care of your valuables especially Cash, Mobile Phones, Laptop and other valuables.
10. Carry your valuables along with you whenever you proceed on leave/vacation.
11. Use electric appliances, lights and water optimally.
12. Keep the campus clean and hygienic.

### **DON'TS**

1. Ragging inside / outside the campus.
2. Possession of Fire arms and daggers etc.
3. Use of Alcohols, Toxic drugs, sheesha, gutkha and hashish/heroin etc.
4. Use of Crackers, explosives and ammunition etc.
5. Smoking and keeping any kind of such items.
6. Misusing college & hostel premises/facilities for activities other than studies.
7. Playing loud music in the room which may disturb studies of colleagues / neighbors.
8. Making noise and raising slogans.
9. Keeping electrical appliances, other than authorized ones.
10. Involvement in politics, ethnic, sectarian and other undesirable activities.
11. Proxy in any manner.
12. Use of mobiles in the classrooms.

- Note:**
1. Rules are revised / reviewed as and when required.
  2. Healthy suggestions are welcome for betterment of Institution